
Advanced Quantum Mechanics - Problem Set 9

Winter Term 2019/20

Due Date: Hand in solutions to problems marked with * before the lecture on **Friday, 20.12.2019, 09:15**. The problem set will be discussed in the tutorials on Wednesday, 08.01.2020, and Friday, 10.01.2020.

24. Addition of angular momenta

3+3+1 Points

Consider two angular momenta $\hat{\mathbf{L}}_1$ and $\hat{\mathbf{L}}_2$ with $l_1 = l_2 = 1$. In this problem we will calculate the eigenvalues and eigenfunctions of $\hat{\mathbf{L}}^2$. The eigenfunctions are linear combinations of the 9 functions

$$Y_{1m}(\theta_1, \varphi_1)Y_{1m'}(\theta_2, \varphi_2) = u_m v_{m'}, \quad \text{with } m, m' = 1, 0, -1.$$

- Construct the 9×9 matrix representation of the operator $\hat{\mathbf{L}}^2$ in the $u_m v_{m'}$ basis.
- Calculate the eigenvalues of $\hat{\mathbf{L}}^2$ by diagonalizing the matrix.
- Calculate the corresponding eigenfunctions.

Hint: It is possible to make the matrix block-diagonal, as shown in the figure, by making suitable row- and column-operations.

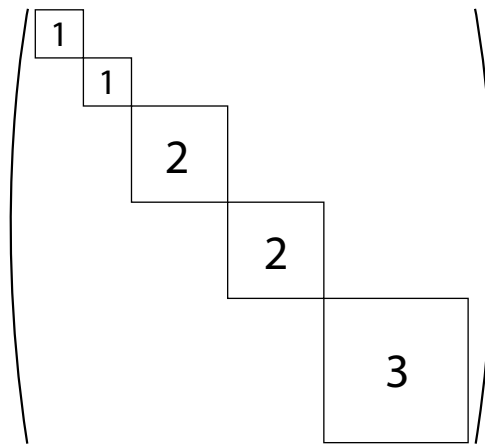


Figure 1: The matrix can be transformed into a block diagonal form.

25. Spin-orbit coupling

2+2+2 Points

Consider a particle with orbital angular momentum $\hat{\mathbf{L}}$ and spin angular momentum $\hat{\mathbf{S}}$. The total angular momentum is $\hat{\mathbf{J}} = \hat{\mathbf{L}} + \hat{\mathbf{S}}$.

- (a) Calculate the expectation value of $\hat{\mathbf{L}} \cdot \hat{\mathbf{S}}$ assuming that the particle is in a state $|l, s; j, m\rangle$.
- (b) An electron is moving in an electrostatic potential $\phi(r)$ with $r = |\mathbf{r}|$. Show that the electric field experienced by the particle is given by

$$\mathbf{E} = -\mathbf{r} \frac{1}{r} \frac{d\phi}{dr}.$$

- (c) In the rest frame of the particle, the particle experiences a magnetic field $\mathbf{B} = -\mathbf{v} \times \mathbf{E}/c^2$. Calculate the energy $\frac{e}{m} \hat{\mathbf{S}} \cdot \mathbf{B}$, where e and m are the electron charge and mass respectively.

Remark: The result found in (c) is off by a factor of two compared to the exact result, which can be obtained using the Dirac equation. The reason is that the simple argument given above assumes a straight-line motion of the particle whereas the potential given above leads to a curved particle trajectory.

*26. Spin-orbit coupling in Hydrogen

4+2+2 Points

The spin-orbit Hamiltonian for Hydrogen is given by

$$H_{\text{SO}} = \frac{e^2}{4\pi\epsilon_0} \frac{1}{m^2 c^2 r^3} \hat{\mathbf{S}} \cdot \hat{\mathbf{L}}.$$

We will treat this Hamiltonian as a perturbation in this problem.

- (a) Using the relevant Hydrogen wave-function, calculate the leading order energy correction due to spin-orbit coupling, for $n = 2$, and $l = 1$. Take $s = 1/2$ as the spin of the electron.
- (b) Use Kramers' relation

$$\frac{\alpha + 1}{n^2} \langle r^\alpha \rangle - (2\alpha + 1)a \langle r^{\alpha-1} \rangle + \frac{\alpha}{4} [(2l + 1)^2 - \alpha^2] a^2 \langle r^{\alpha-2} \rangle = 0,$$

where a is the Bohr radius, to derive a relation between $\langle r^{-2} \rangle$ and $\langle r^{-3} \rangle$.

- (c) Calculate the leading order energy correction due to spin-orbit coupling for general n and l . You may use that

$$\langle r^{-2} \rangle = \frac{1}{(l + 1/2)n^3 a^2}.$$

Hint: The result of task 25 (a) might be helpful.