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## Quantum Mechanics 2- Problem Set 9

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*Wintersemester 2016/2017*

**Abgabe:** The problem set will be discussed in the tutorials on **Thursday, 15.12.2016, 09:15**  
and **Friday, 16.12.2016, 11:15**

### 26. Perturbation theory

*2+2+2 Punkte*

In this problem we will estimate the polarizability of the Hydrogen atom in its ground state  $|0\rangle$ . In the presence of an electric field  $E$  in the  $z$ -direction the Hydrogen Hamiltonian is perturbed by  $eEz$ .

- (a) Show that the energy shift in the ground state is, to leading order,

$$\Delta E = e^2 E^2 \sum_{k \neq 0} \frac{|\langle k|z|0\rangle|^2}{E_0 - E_k}.$$

- (b) Use your result in (a) to show that the polarizability is given by

$$\alpha = \frac{2e^2}{\epsilon_0} \sum_{k \neq 0} \frac{|\langle k|z|0\rangle|^2}{E_k - E_0}.$$

*Hint: The energy induced by a dipole  $d$  is  $-(1/2)dE^2$*

- (c) Use that  $E_k \geq E_1$ , where  $E_1$  is the energy of the first excited state, and that

$$\sum_{k \neq 0} \langle 0|z|k\rangle \langle k|z|0\rangle = \sum_k \langle 0|z|k\rangle \langle k|z|0\rangle - \langle 0|z|0\rangle^2 = \langle 0|z^2|0\rangle,$$

to derive an upper bound on  $\alpha$ . Compare this with the experimental value  $\alpha = 8.5 \cdot 10^{-30} \text{m}^3$ .

### 27. Spin-orbit coupling

*2+2+2 Punkte*

Consider a particle with orbital angular momentum  $\mathbf{L}$  and spin angular momentum  $\mathbf{S}$ . The total angular momentum is  $\mathbf{J} = \mathbf{L} + \mathbf{S}$ .

- (a) Treating the angular momentum operators semi-classically as vectors, calculate the expectation value of  $\mathbf{L} \cdot \mathbf{S}$ .

- (b) An electron is moving in an electrostatic potential  $\phi(r)$ . Show that the electric field experienced by the particle is given by

$$\mathbf{E} = -\mathbf{r} \frac{1}{r} \frac{d\phi}{dr}.$$

- (c) In the rest frame of the particle, the particle experiences a magnetic field  $\mathbf{B} = -\mathbf{v} \times \mathbf{E}/c^2$ . Calculate the interaction energy  $\frac{e}{m} \mathbf{S} \cdot \mathbf{B}$ , where  $e$  and  $m$  are the electron charge and mass respectively. Your result should be off by a factor of two compared to the exact result obtained using the Dirac equation. Why?

## 28. Spin-orbit coupling in Hydrogen

*4+2+2 Punkte*

The spin-orbit Hamiltonian for Hydrogen is given by

$$H_{\text{SO}} = \frac{e^2}{4\pi\epsilon_0} \frac{1}{m^2 c^2 r^3} \mathbf{S} \cdot \mathbf{L}.$$

We will treat this Hamiltonian as a perturbation in this problem.

- (a) Using the relevant Hydrogen wave-function, calculate the leading order energy correction due to spin-orbit coupling, for  $n = 2$ , and  $l = 1$ . Take  $s = 1/2$  as the spin of the electron.
- (b) Use Kramers' relation

$$\frac{\alpha + 1}{n^2} \langle r^\alpha \rangle - (2\alpha + 1)a \langle r^{\alpha-1} \rangle + \frac{\alpha}{4} [(2l + 1)^2 - \alpha^2] a^2 \langle r^{\alpha-2} \rangle = 0,$$

where  $a$  is the Bohr radius, to derive a relation between  $\langle r^{-2} \rangle$  and  $\langle r^{-3} \rangle$ .

- (c) Calculate the leading order energy correction due to spin-orbit coupling for general  $n$  and  $l$ . You may use that

$$\langle r^{-2} \rangle = \frac{1}{(l + 1/2)n^3 a^2}.$$