



Materialaspekte multikristalliner Solarzellen aus umg-Si-Feedstock



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FAHL Academia, Oppurg, 29. September 09

Q.CELLS



Overview

- **Introduction: Impurity concentration in different Silicon Qualities**
- **Production of multicrystalline Ingots and Wafers**
 - Ingot Growth
 - Sawing process
 - Segregation – an important process
- **Companies that produce umg-Si / SOG-Si and their processes**
 - Elkem
 - BSI
 - 6N Silicon
- **Influence of impurities**
 - Dopants
 - Metals
 - Alkaline Metals
 - O / C / N
- **Summary and conclusions**

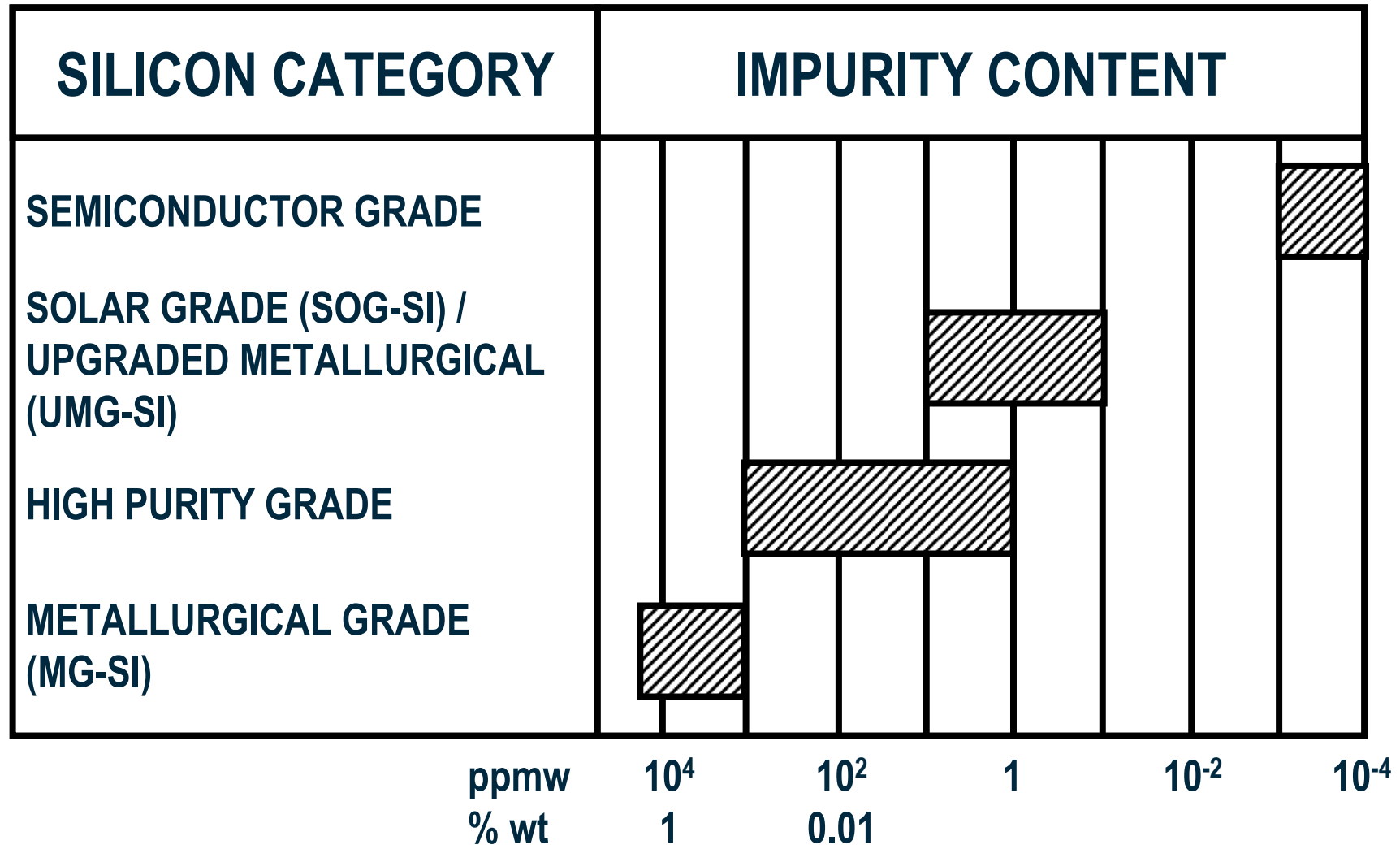


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Impurity concentration in different Silicon Qualities





Semiconductor vs. Solar Grade Silicon

Advantages of Semiconductor Grade Silicon:

- Processing from Feedstock to Cell is well known
- Quality is high enough for all high efficiency cell concepts

Advantages of SOG-SI / UMG-SI:

- Energy consumption and payback time lower (→ „greener“ product)
- Lower production and investment costs
- Ramp up of factories and capacities faster → Reaction time to changes in market conditions shorter
- „second source“ to Semiconductor Grade Silicon

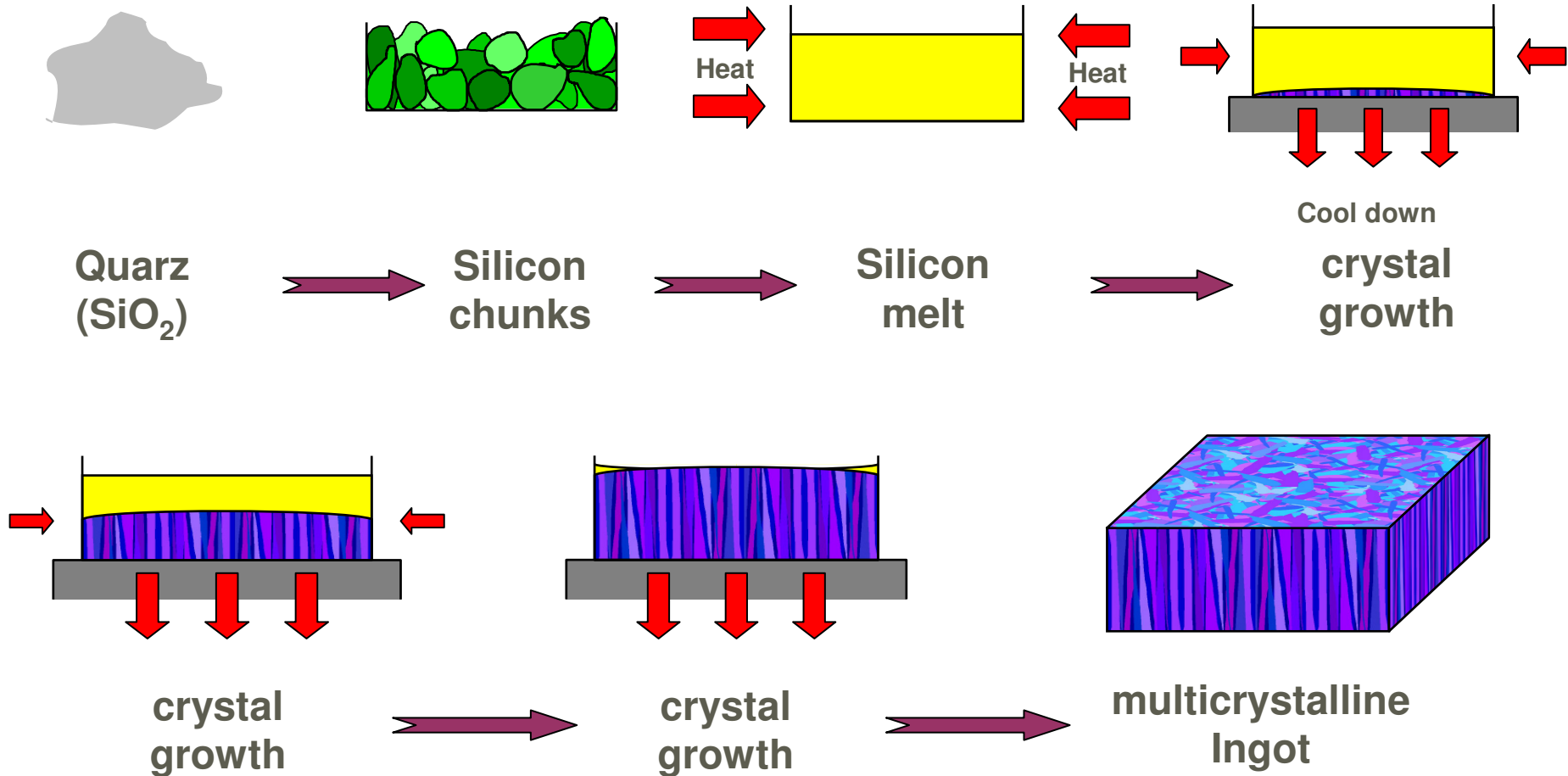


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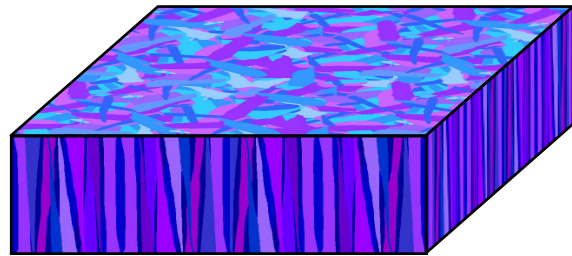


Crystallisation: multicrystalline Ingot

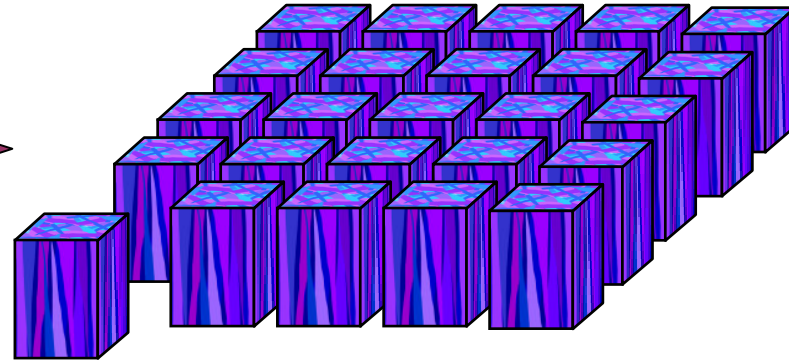




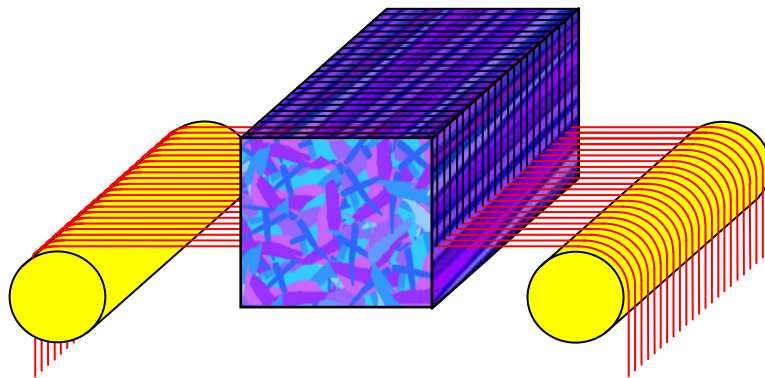
Wire - sawing: multicrystalline wafers



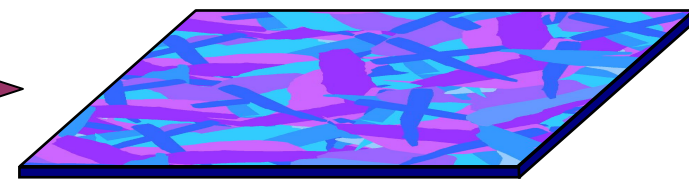
Ingot
~ 80 x 80 x 30 cm
400 kg



Sawing into bricks



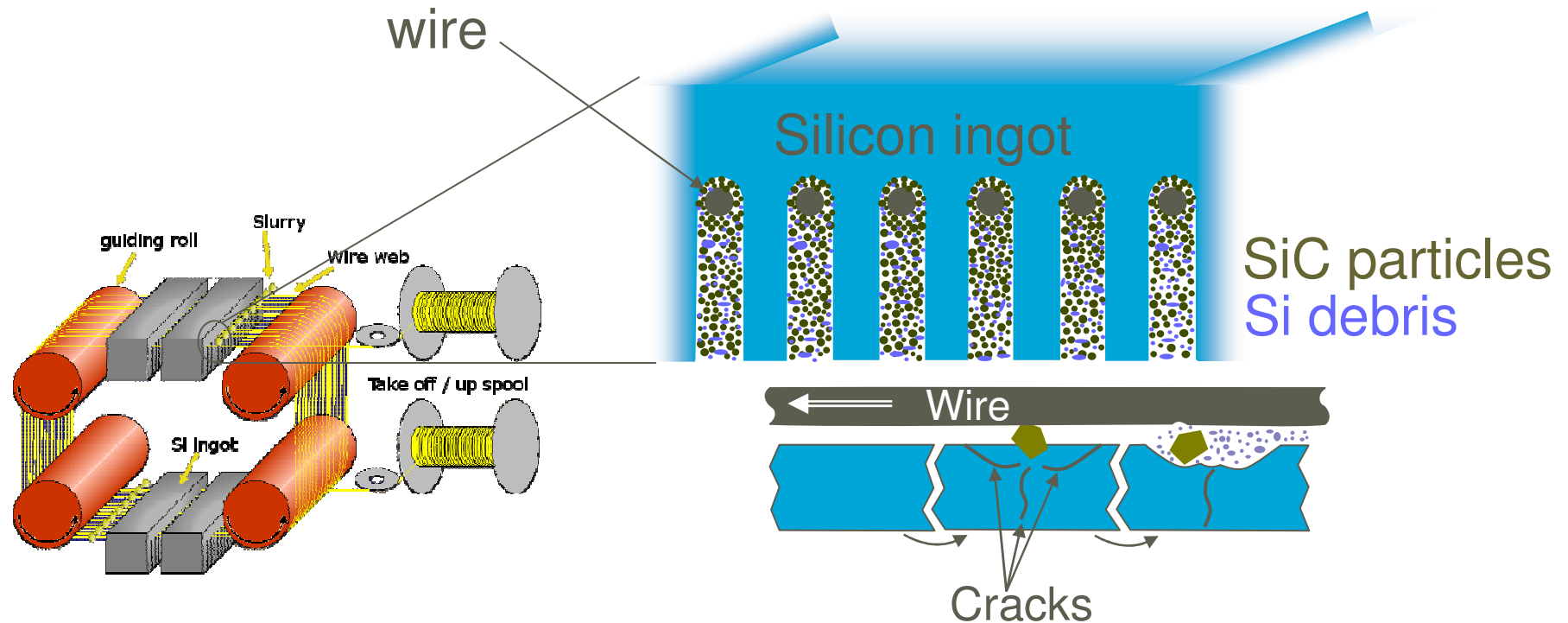
Sawing into wafers
per brick ~ 600 wafers



Multicrystalline Silicon Wafer
156 x 156 mm
Thickness 180 μm



The principle of wire-sawing



- A steel wire is wound around wire-guides
- Some Slurry (PEG + SiC) is poured on the wires
- The SiC particles wear the silicon ingot out → slices wafers
- This process induces cracks in the wafers → they become brittle



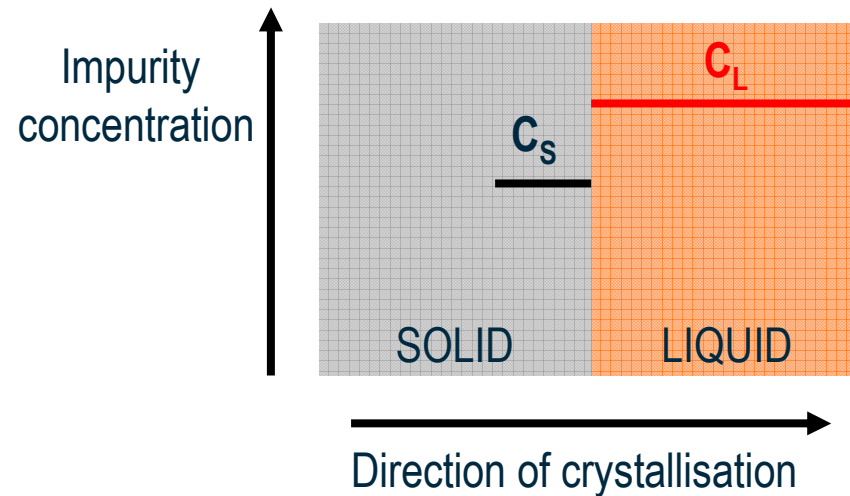
Segregation – a central process

At the growth front, the concentration incorporated in the crystallised phase is different to the concentration in the melt

The Segregation coefficient k is defined as:

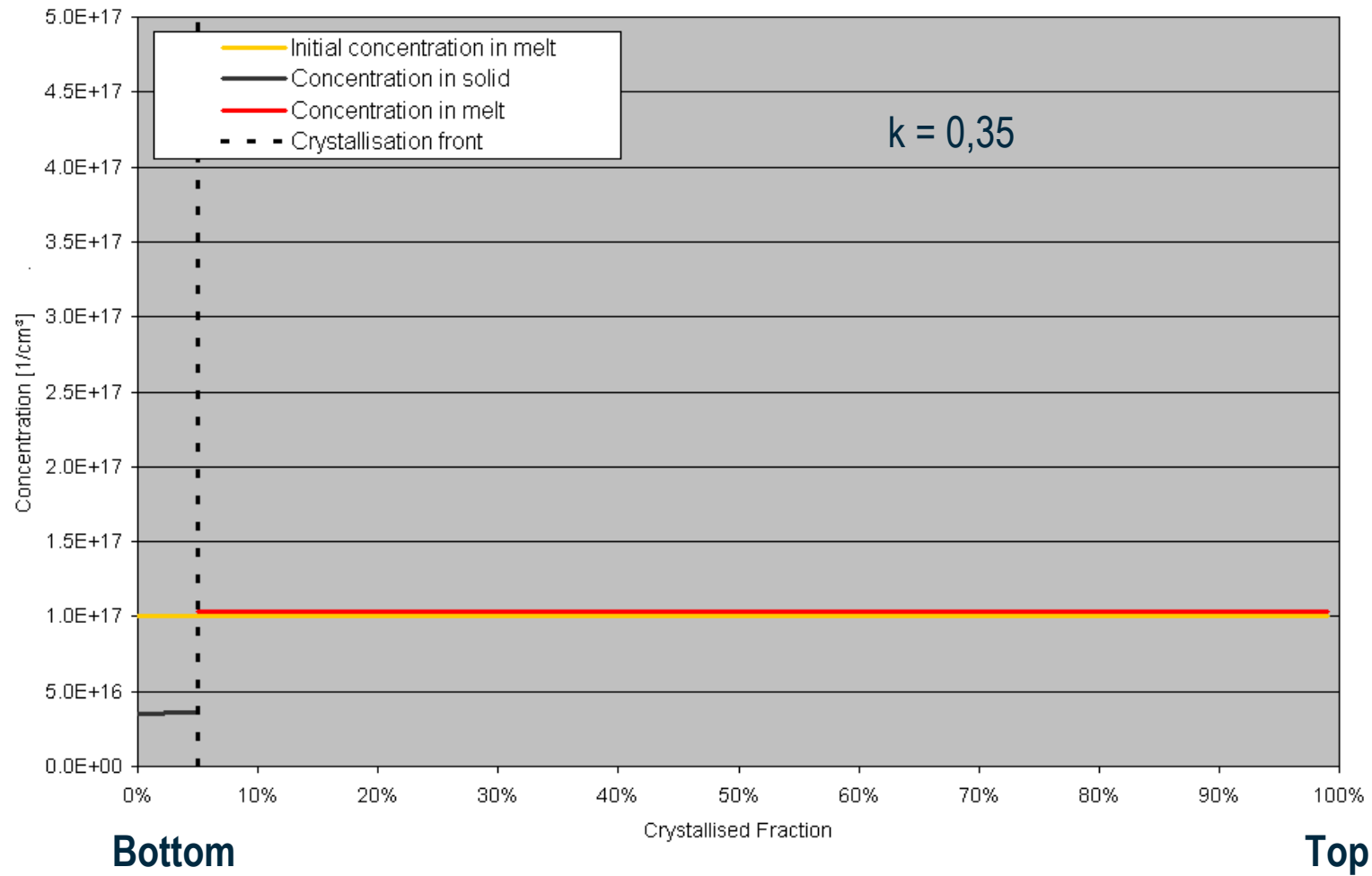
$$k = \frac{C_S}{C_L}$$

C_L concentration in the melt
 C_S concentration in the solid



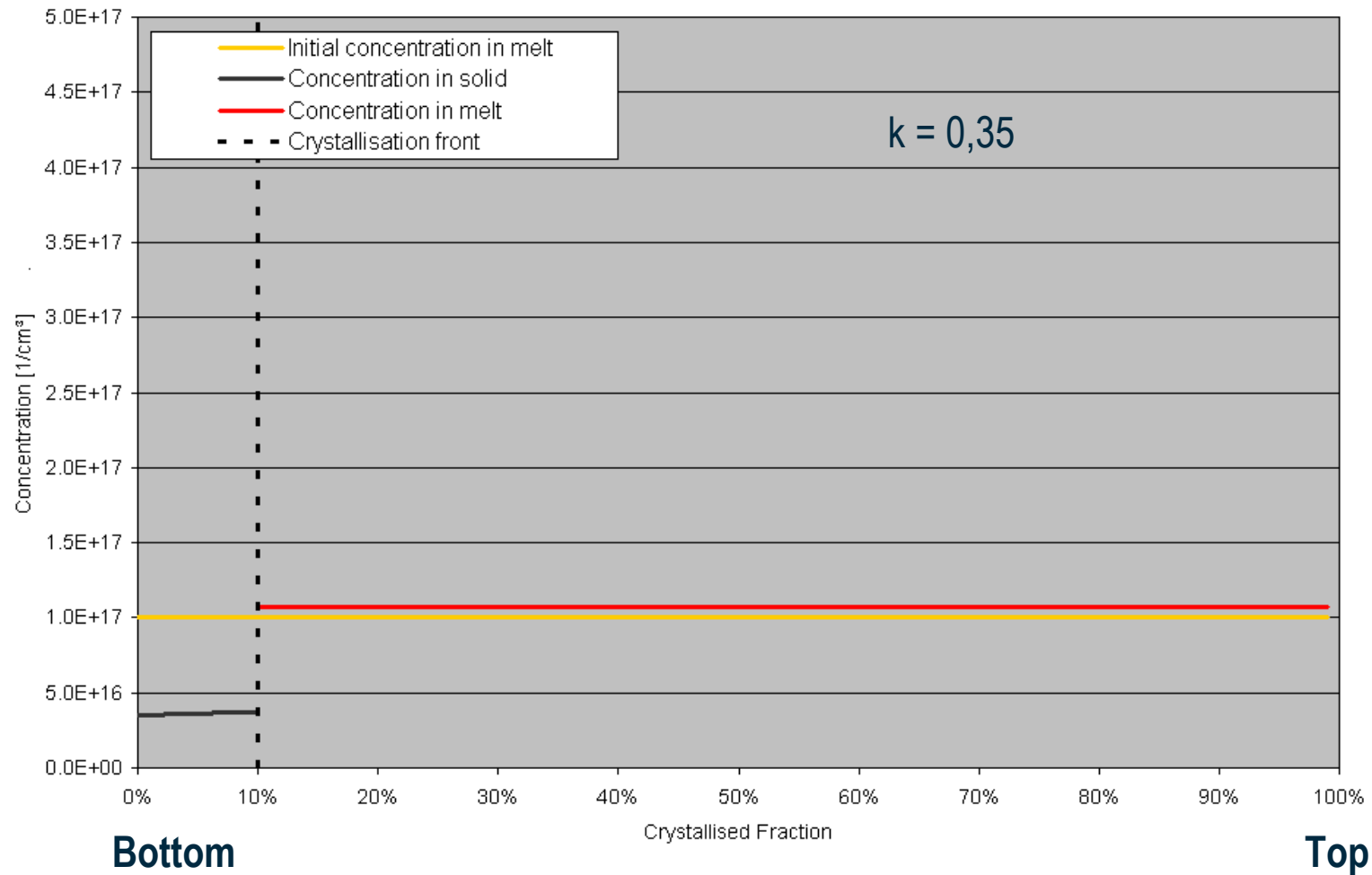


Segregation



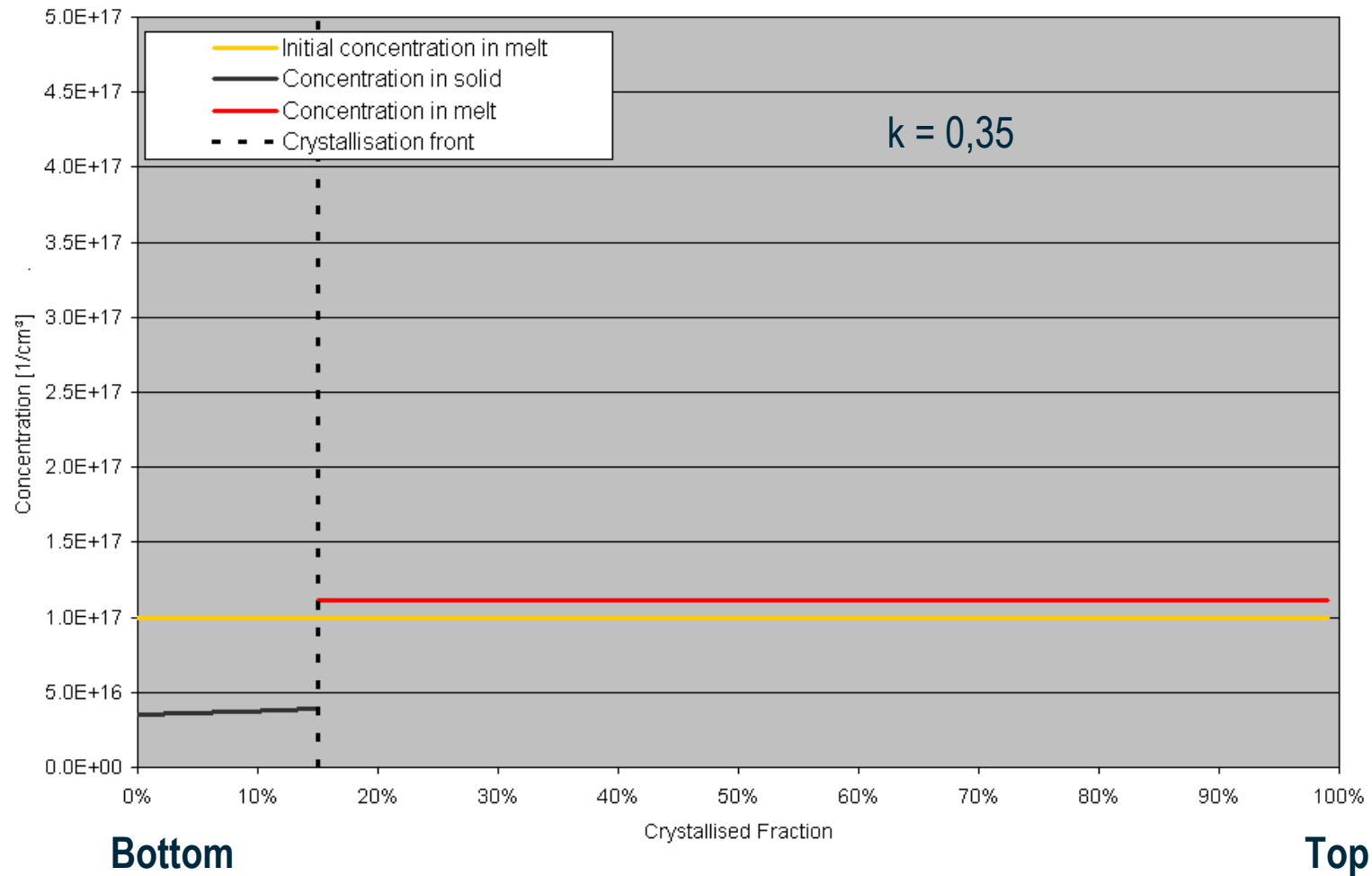


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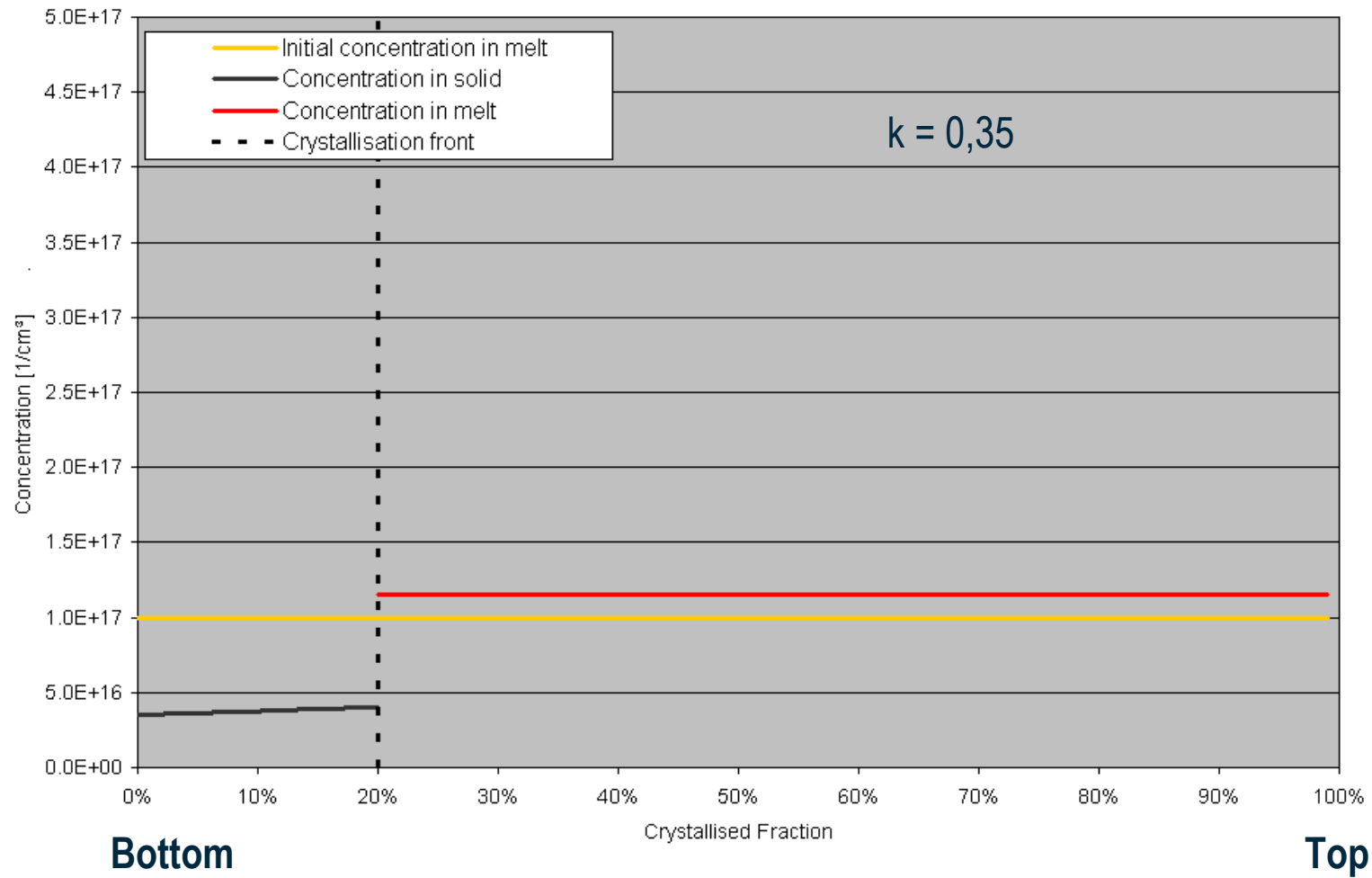


Segregation



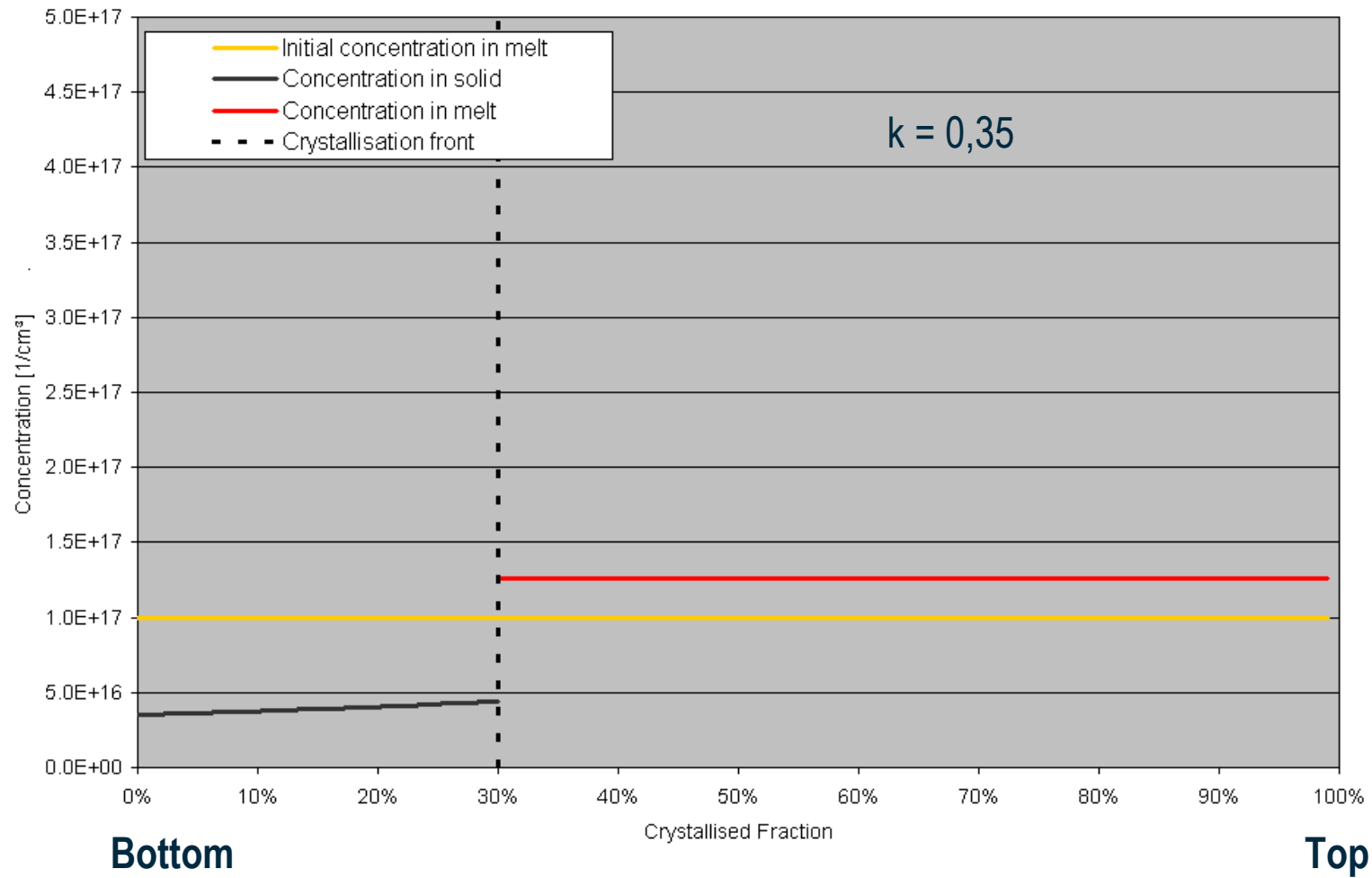


Segregation



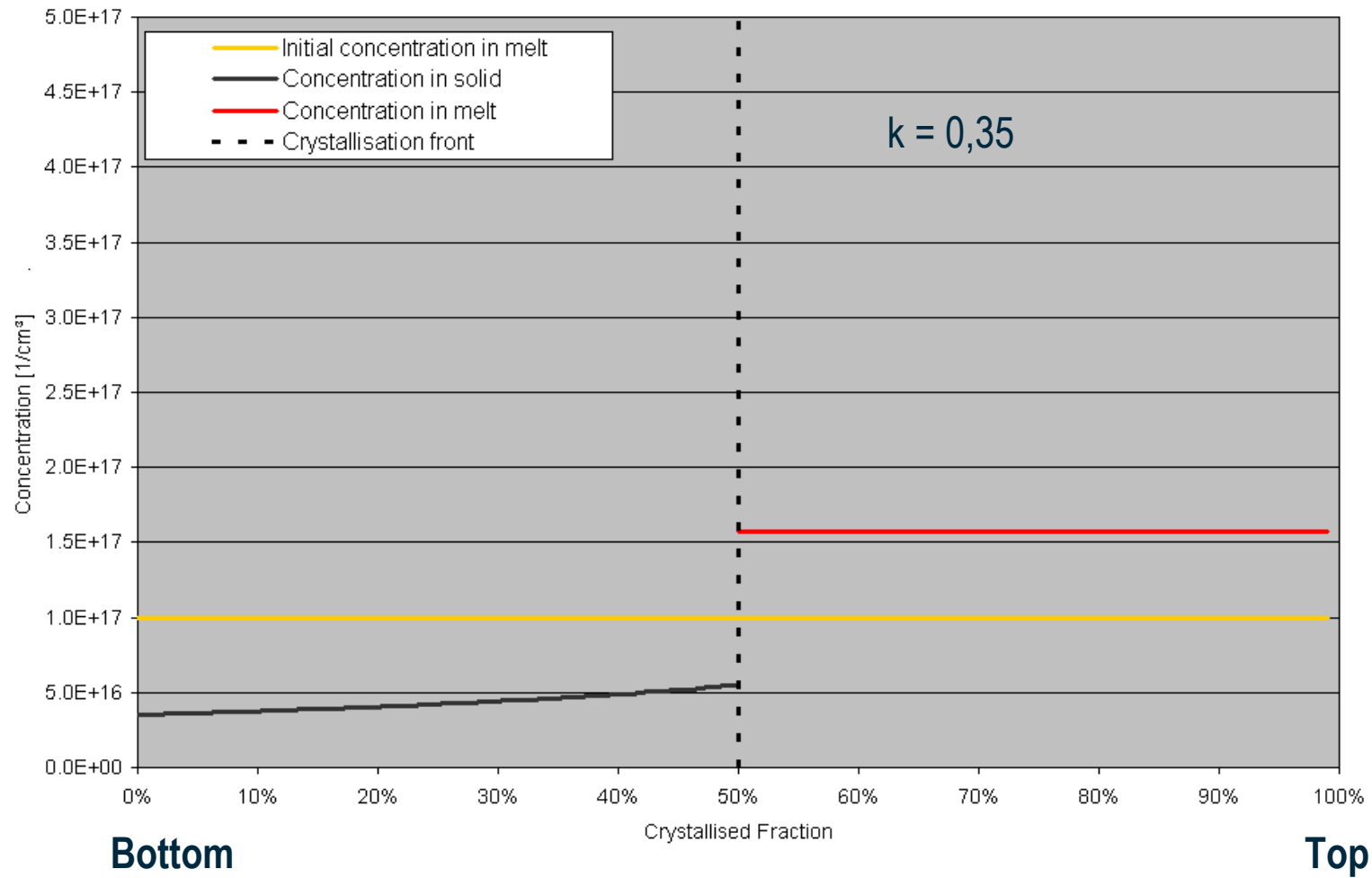


Segregation



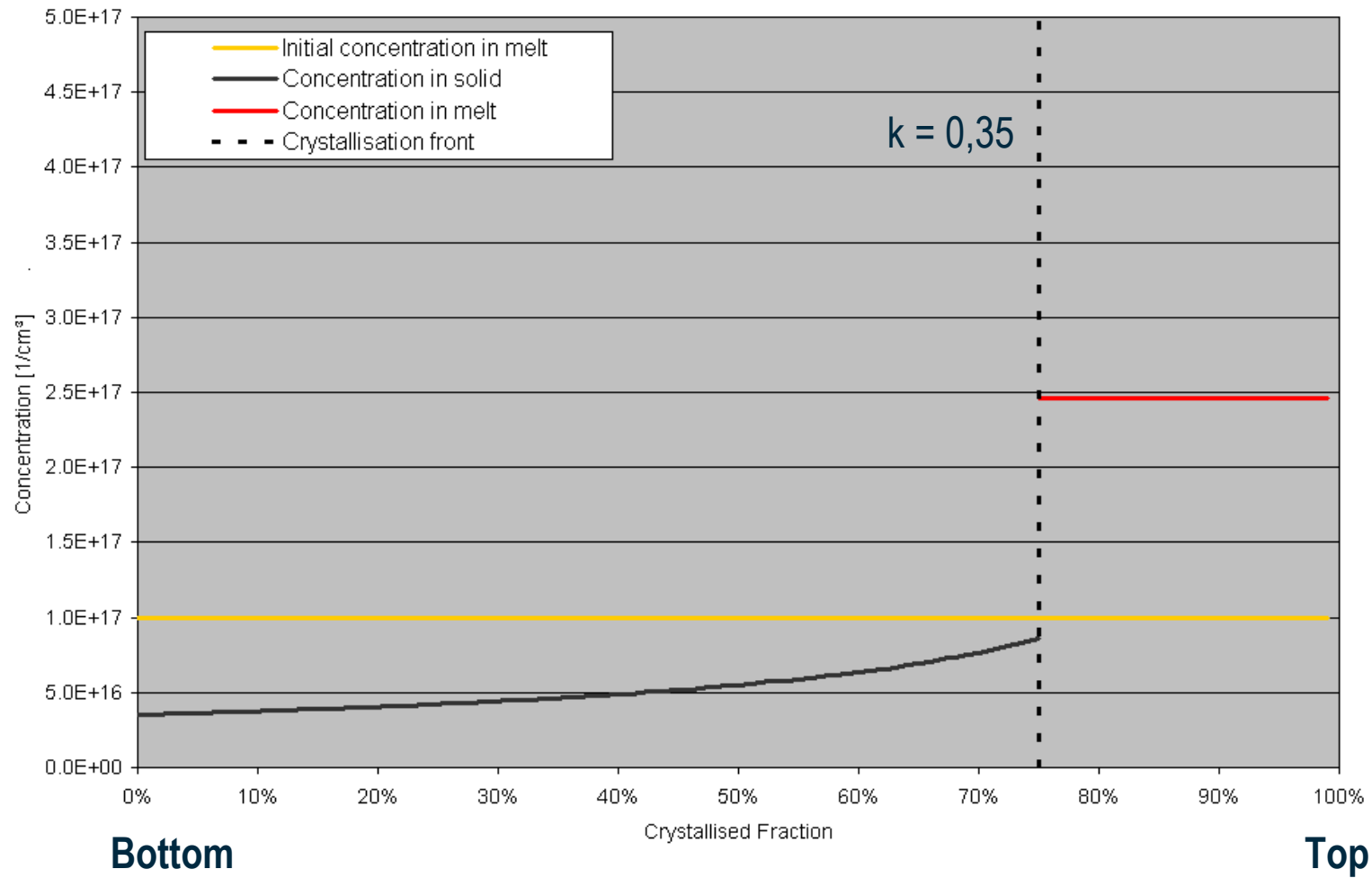


Segregation



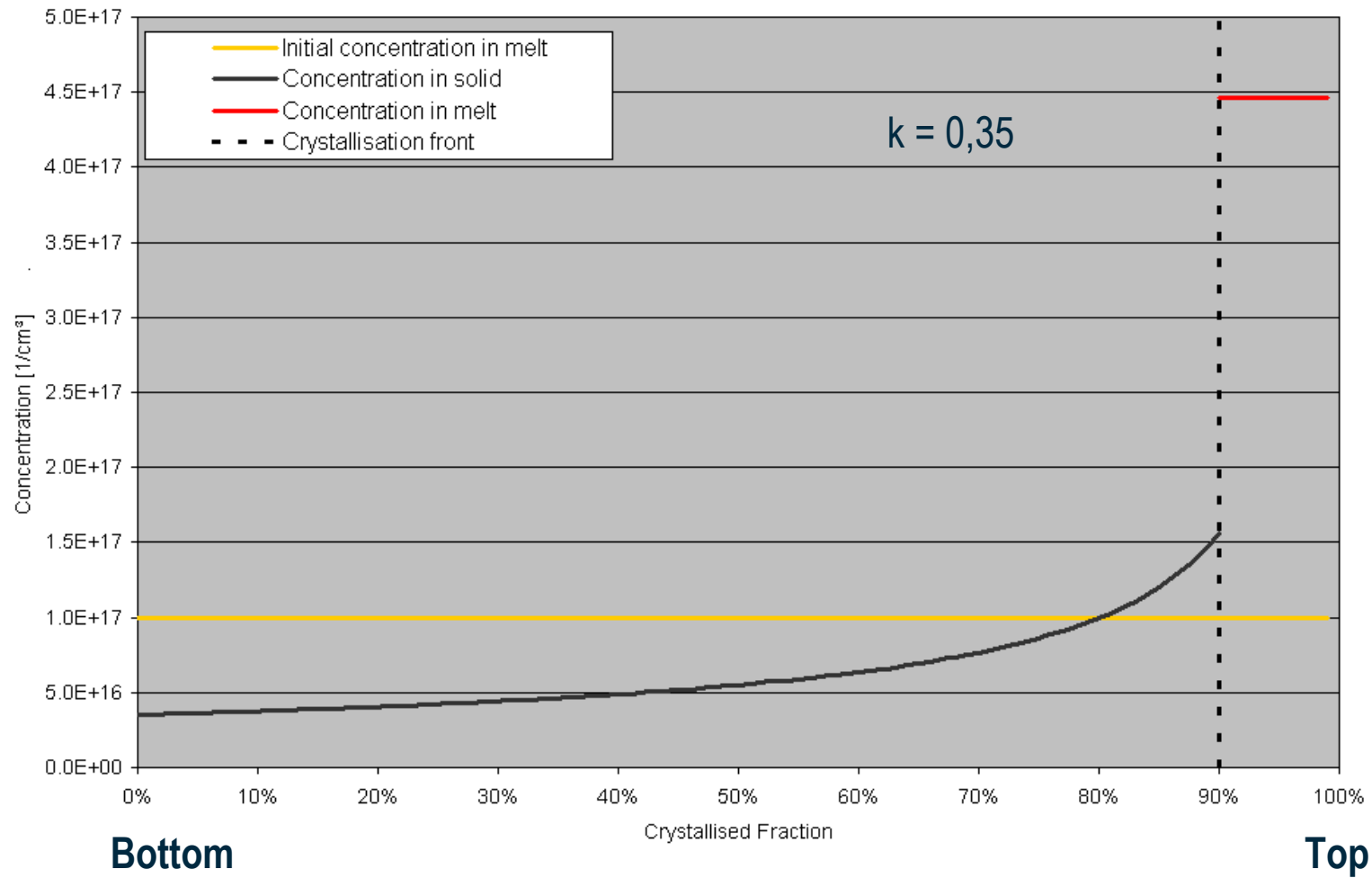


Segregation



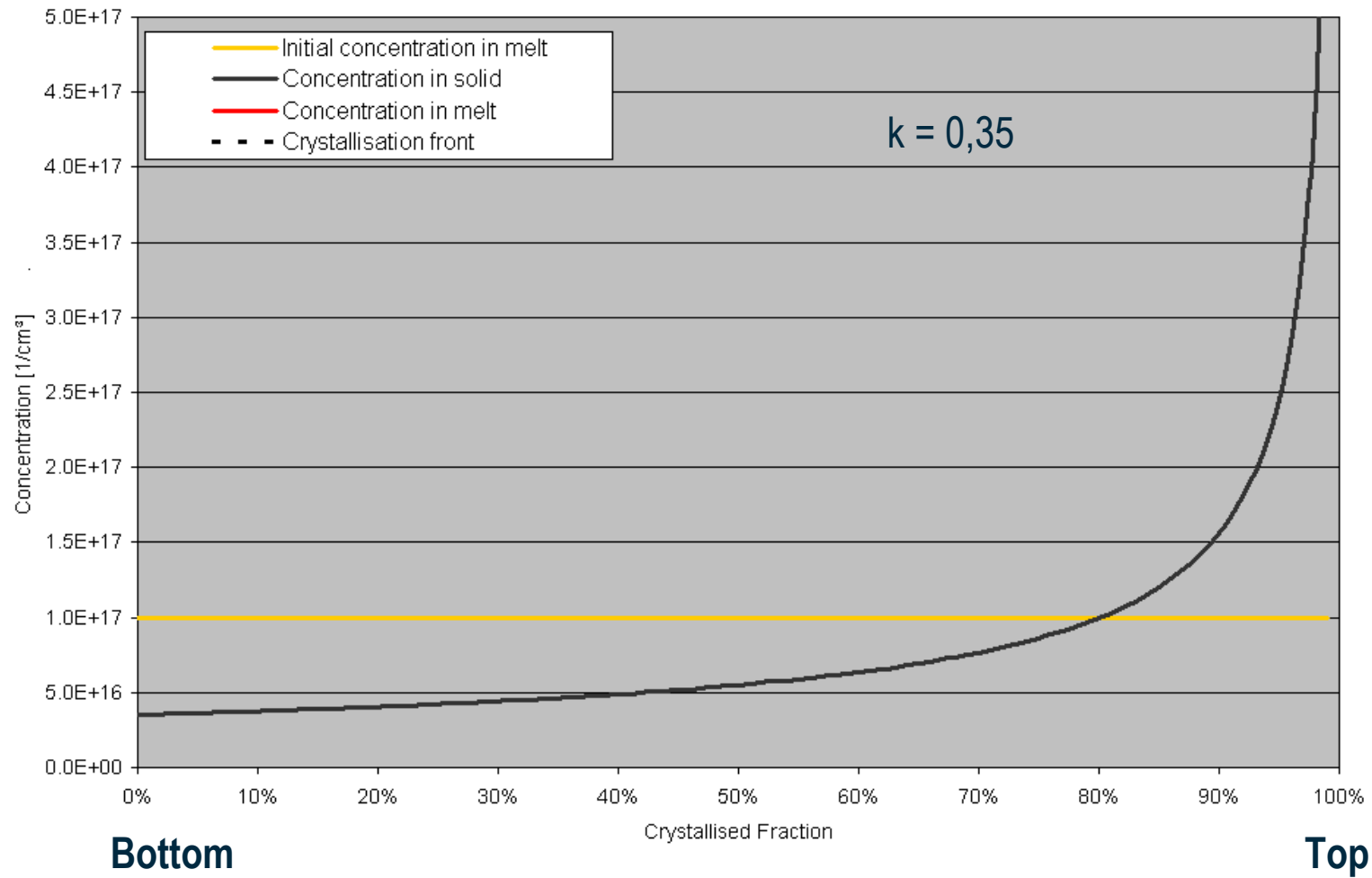


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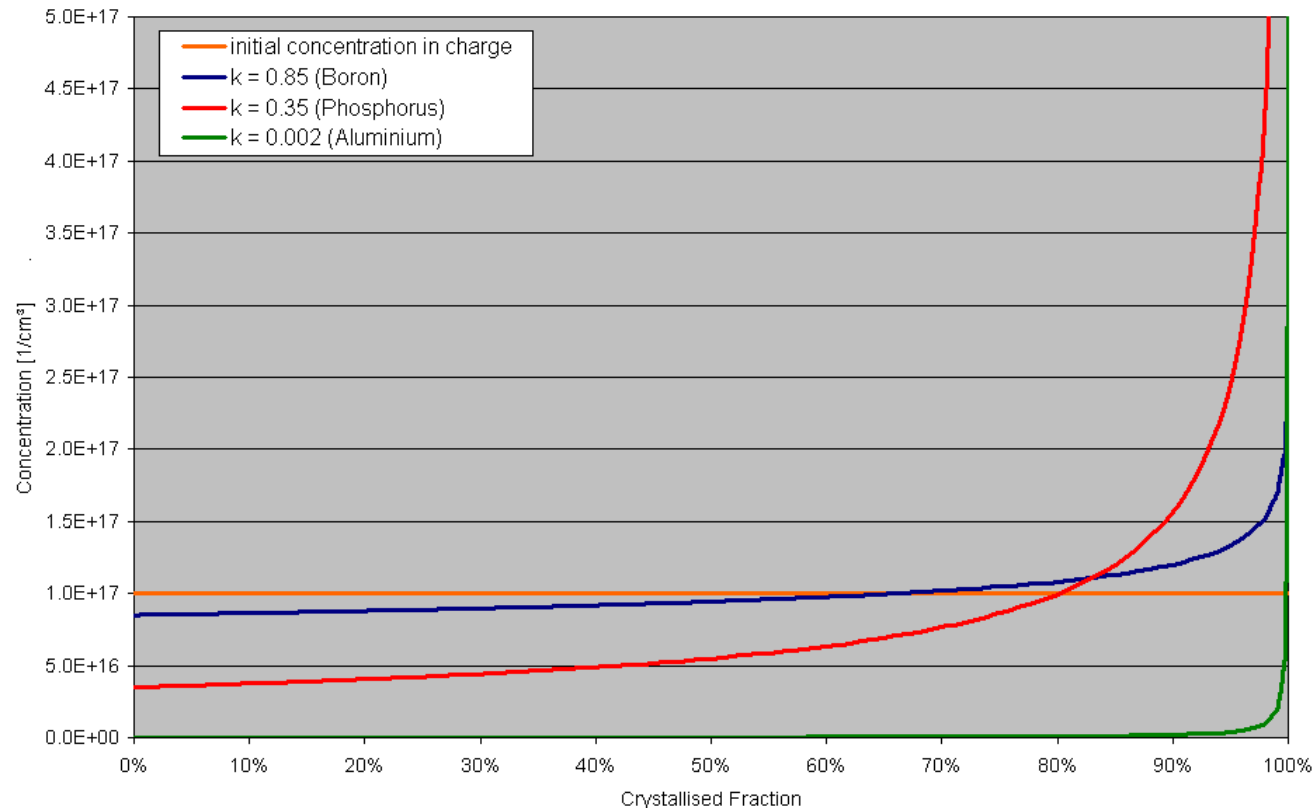


Segregation





Cleaning by solidification ?



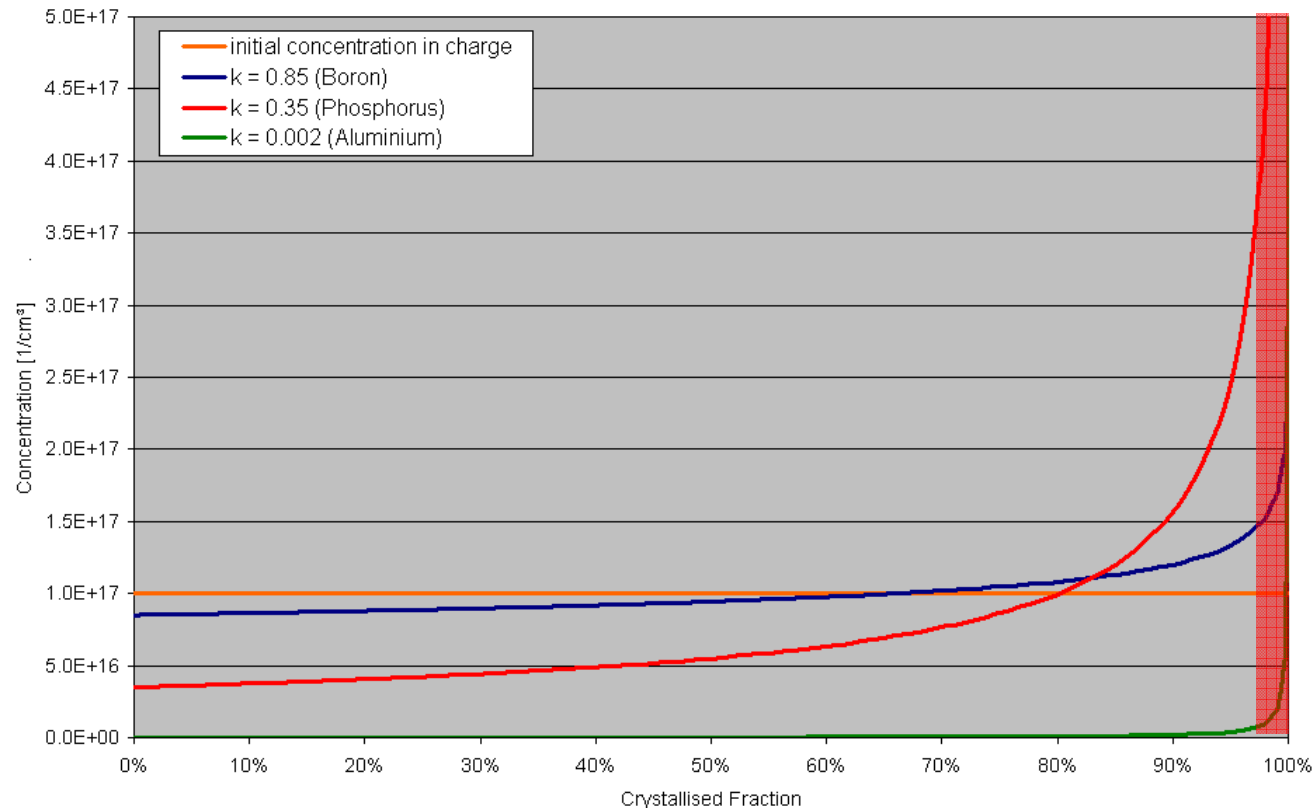
Elements are distributed non uniformly in the crystal.

$k \sim 1 \rightarrow$ distribution is relatively homogenous

$k \ll 1 \rightarrow$ elements segregate towards the top of the ingot



Cleaning by solidification ?



Rejecting a small amount of material from the top

- $k \sim 1$ (Dopants like B / P) → no strong cleaning effect

- $k \ll 1$ (e.g. metals) → efficient cleaning

→ **Biggest task for all umg-Si producers is the reduction of B and P**



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Process Elkem Solar



Reduction of
dopants

Reduction of
metals



- In-house production only
- Based on Elkem's core competencies

- Three sequential purification steps designed to reduce the level of impurities for critical elements
- Largely based on Elkem's core competencies in high temperature processes, process and equipment design

- Ingots cleaned and sawed into bricks of ~10 kg
- Quality control

Elkem Solar – status and future outlook, Helge Aasen, 6th Solar Silicon Conference, Munich, 2008



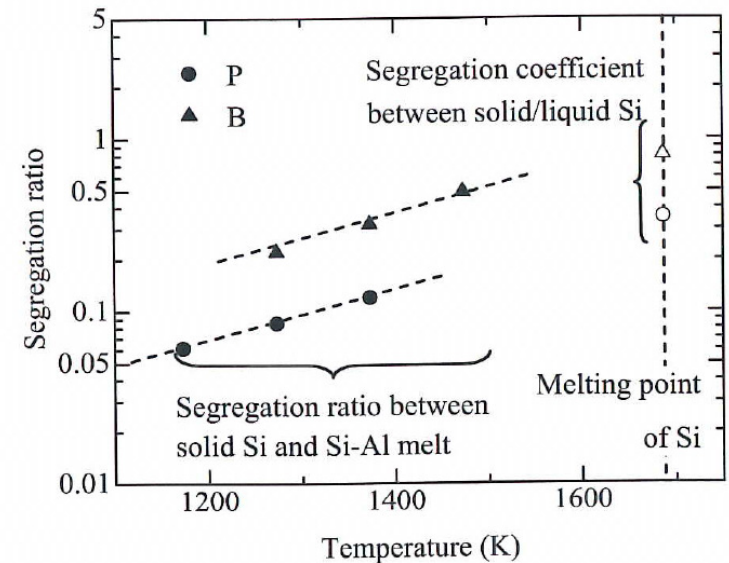
Process 6N Silicon



Aluminum/Silicon Melt

- Lower temperature than molten Silicon
- Safer
- Less expensive, lower energy costs
- Fast, efficient purification

Solar Silicon Solution in a Dynamic Market!, David Dunnison, 7th Solar Silicon Conference, Munich, 2009



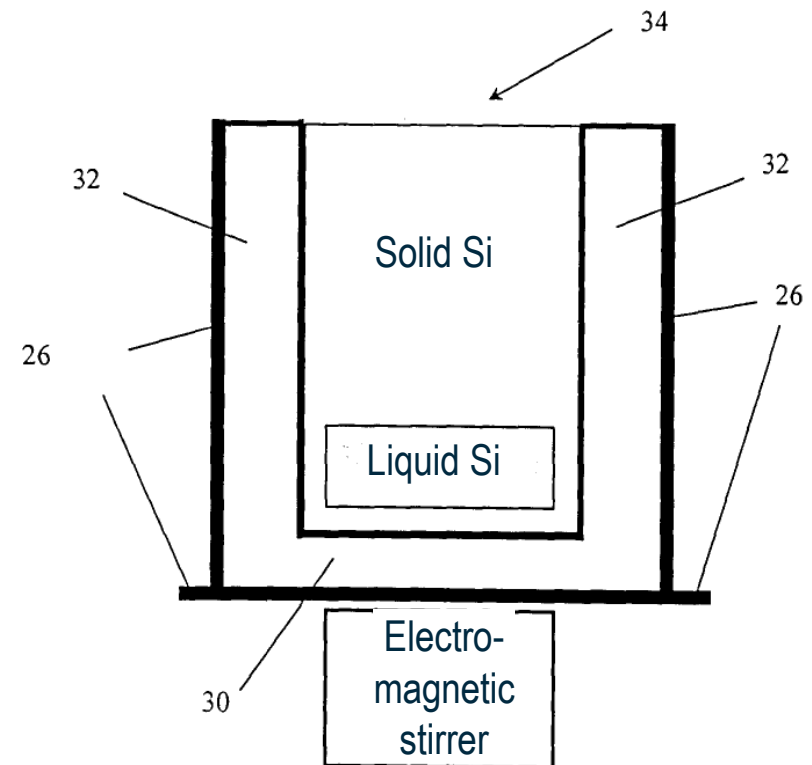
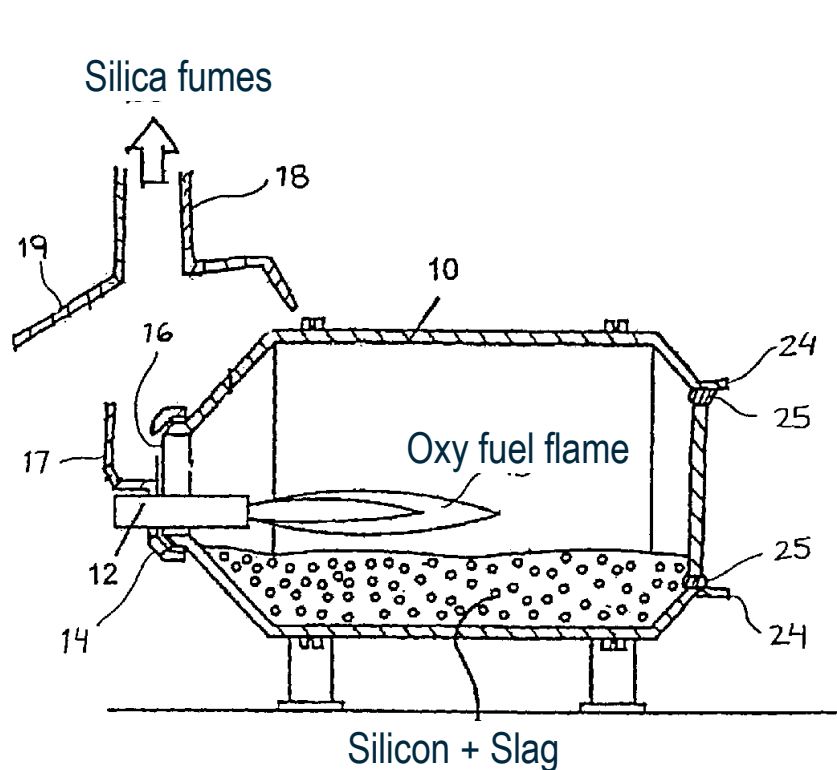
Low temperature solidification refining of solar grade silicon using Si-Al solvent, K. Moriata, Silicon for the Chemical and Solar Industry, Oslo, 2008



Process BSI Becancours / Timminco

Rotary furnace with slag and oxidising flame
→ Boron reduction

low cost solidification
→ reduction of metals



Patent WO08031229_2008_BSI_process and Apparatus for purifying Low-Grade silicon material



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Impurities in Silicon

Typical influences of impurities in Silicon

Impurity	Examples	Influence
Dopants	B, P, Ga	<ul style="list-style-type: none">▪ Resistivity, Yield▪ Light Induced Degradation
Metals	Fe, Co, Ni	<ul style="list-style-type: none">▪ Defectlevels in Middle of Bandgap → Recombination
Alkali-Metals	Ka, Na	<ul style="list-style-type: none">▪ Crystallisation: Corrosion of Crucible
Non Metals	C, N, O	<ul style="list-style-type: none">▪ Inclusions▪ Light Induced Degradation▪ Wire breakage▪ shunts



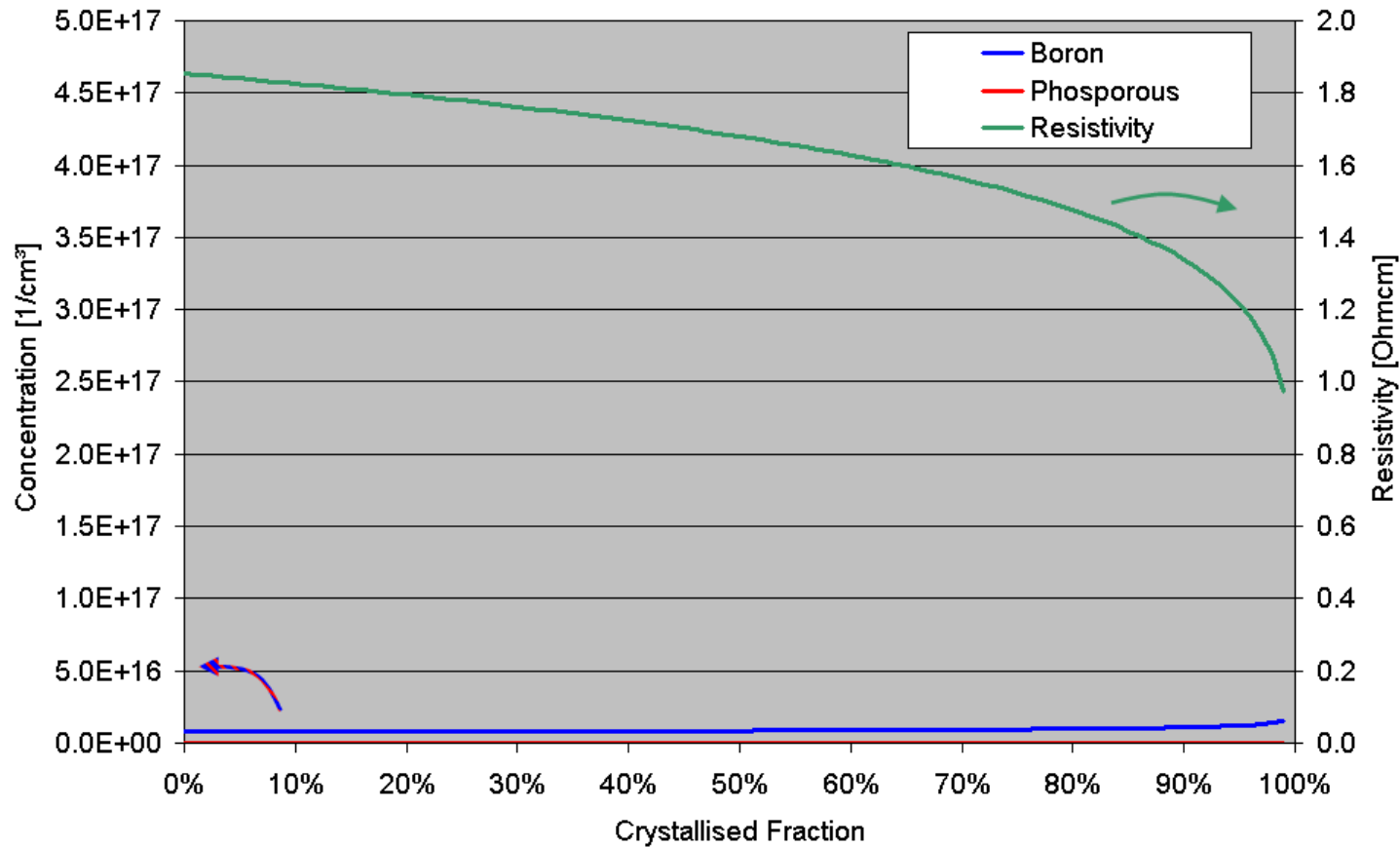
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Influence of Dopants - Resistivity

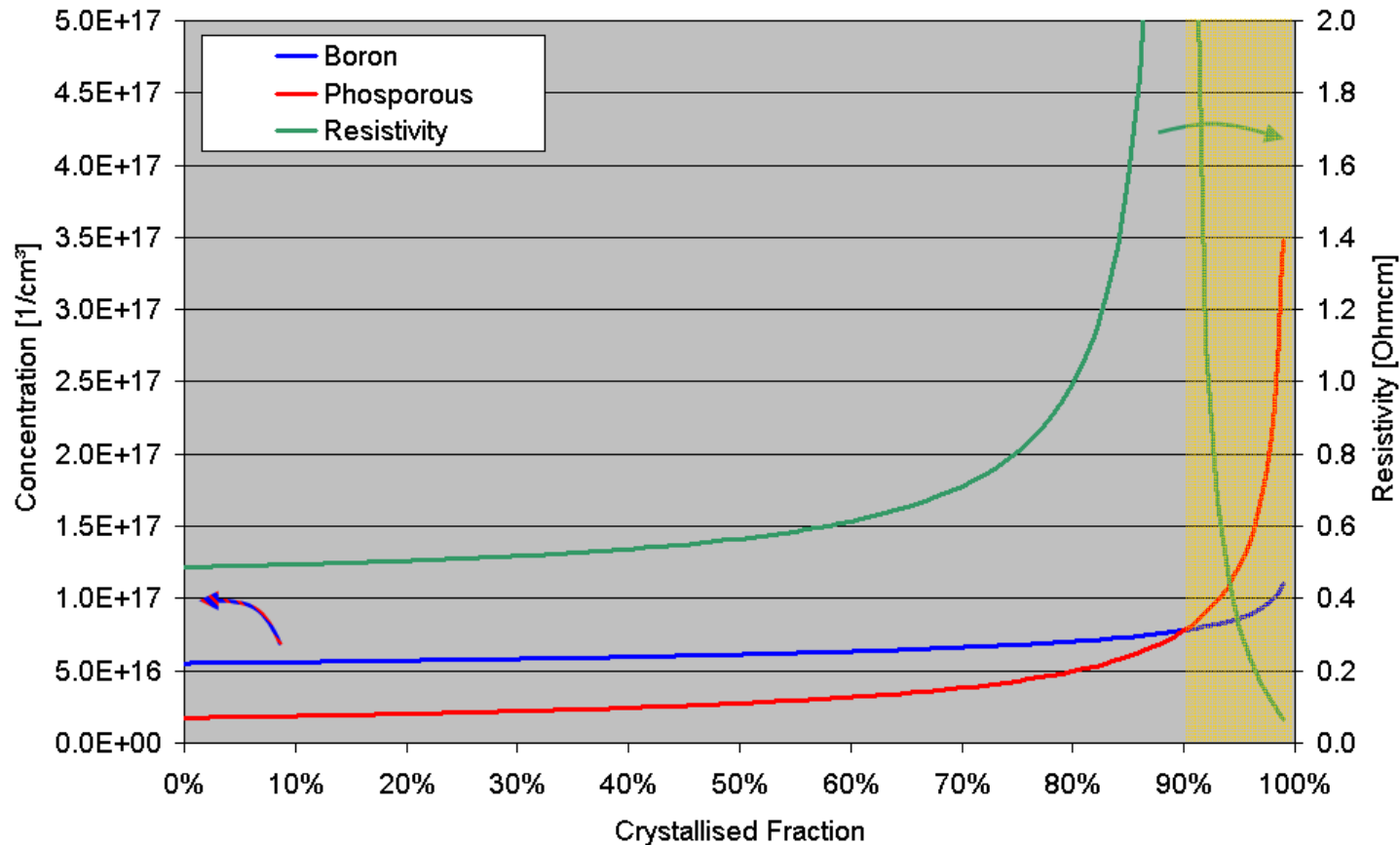


Feedstock: semiconductor grade Silicon

0.07 ppmw B intentionally added to the feedstock → resistivity distribution ~ 1-2 Ωcm, optimal range for standard cell concepts



Influence of Dopants - Resistivity



Feedstock: umg-Si

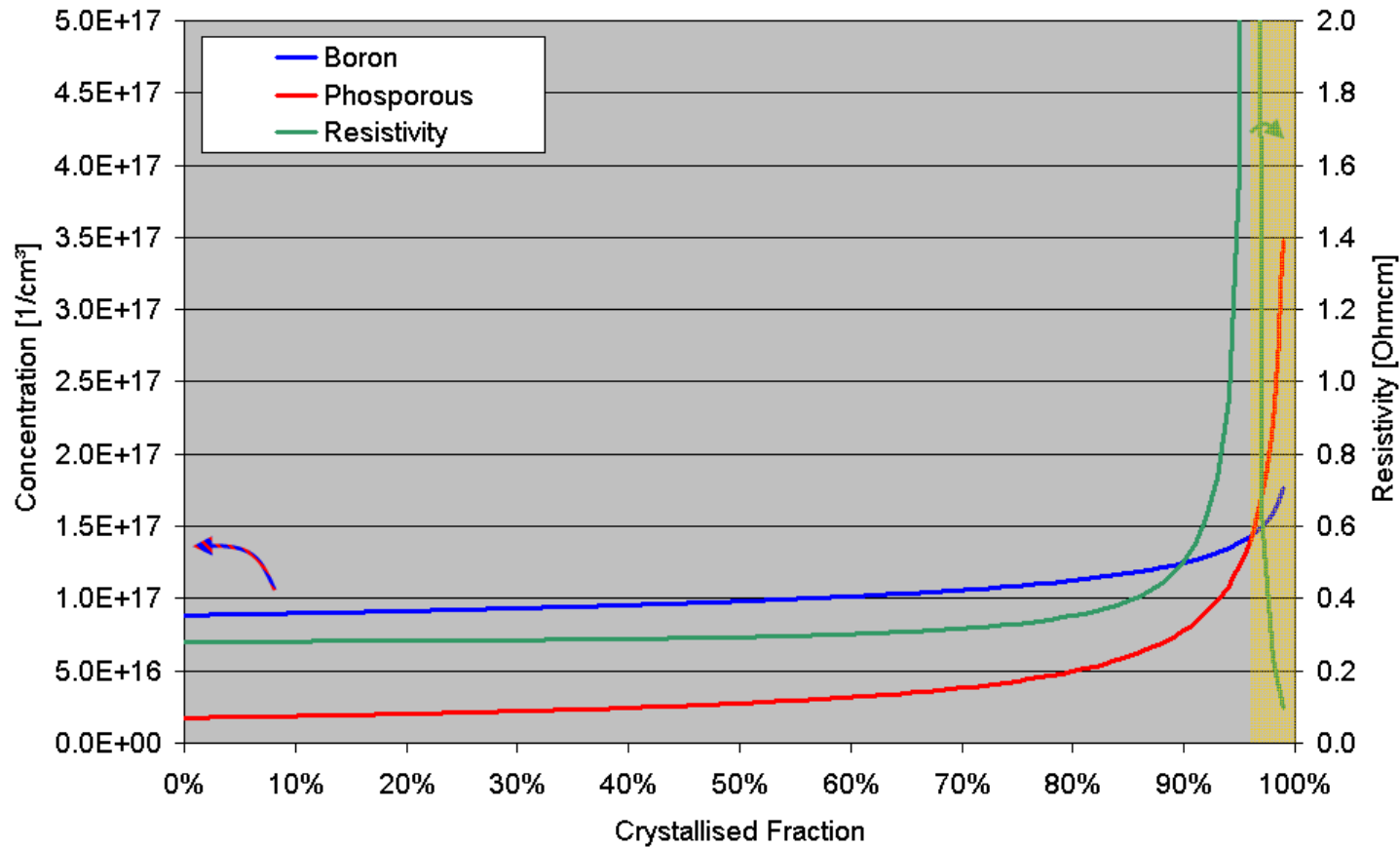
~ 0.5 ppmw of B and 1.1 ppmw of P,

→ resistivity distribution > 0,5 Ohmcm → small efficiency losses

→ Yield loss due to transition to n-type material in top of Ingot



Influence of Dopants - Resistivity



Feedstock: umg-Si + Addition of Boron

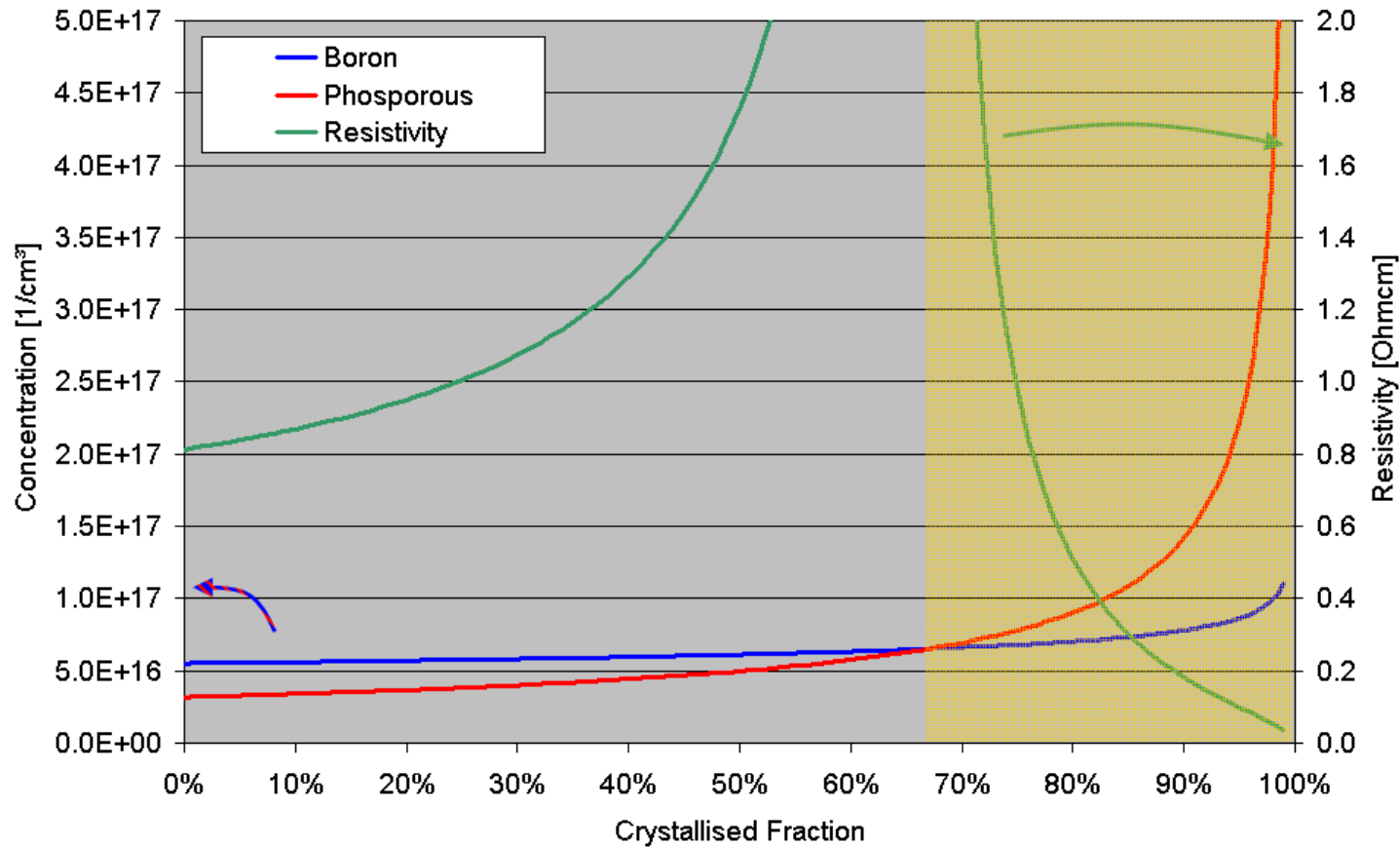
~ 0.8 ppmw of B and 1.1 ppmw of P,

→ Yield increases

→ Average Resistivity decreases → efficiency decreases



Influence of Dopants - Resistivity



Feedstock: umg-Si + Addition of Phosphorus

~ 0.5 ppmw of B and 2.0 ppmw of P

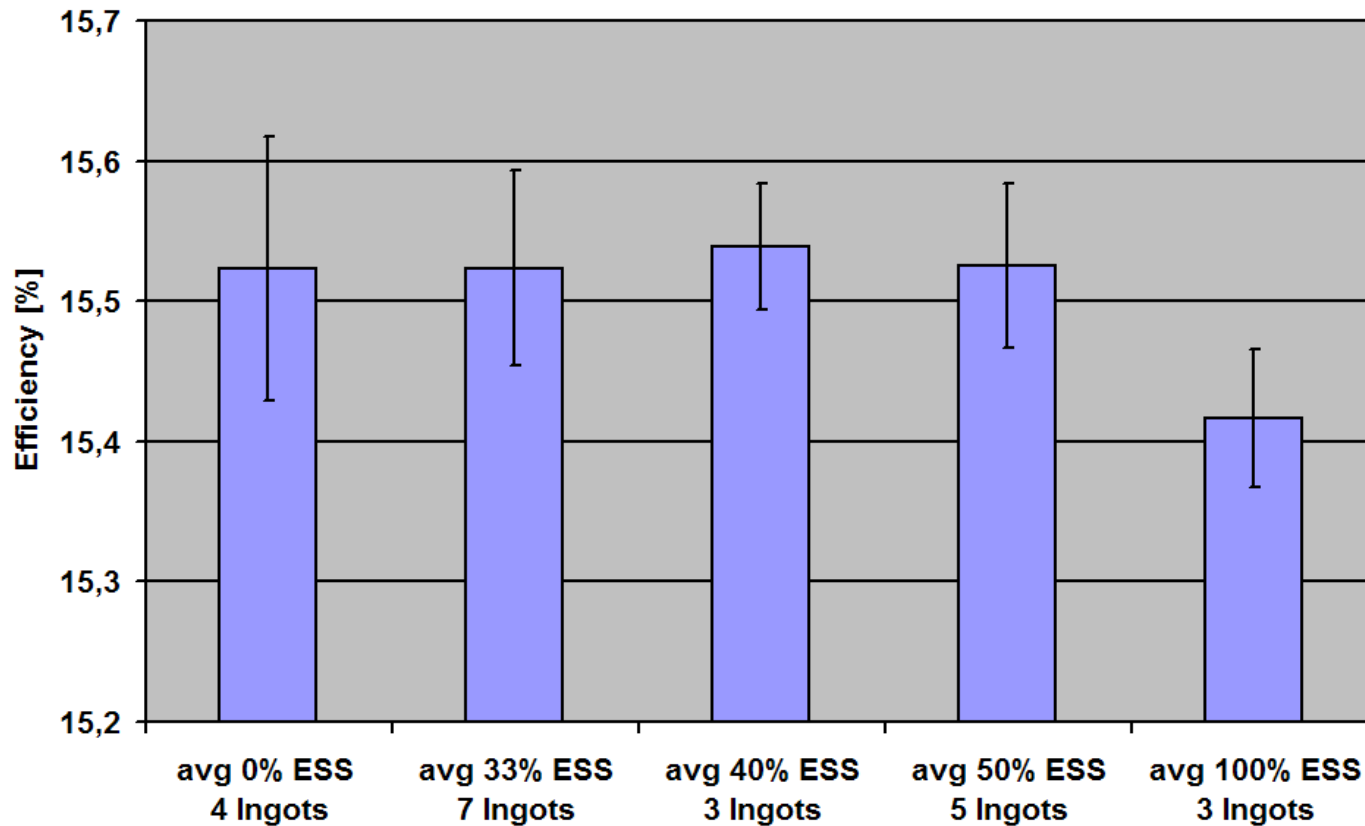
→ Average Resistivity increases → efficiency rises

→ Yield decreases



Results on cell efficiency with Elkem umg-Si

average cell efficiencies for 18 ESS ingots



FIRST RESULTS ON INDUSTRIALIZATION OF ELKEM SOLAR SILICON AT PILLAR JSC AND Q-CELLS
Volker Hoffmann, PVSEC, Valencia, 2008



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Metallic impurities can limit cell efficiency

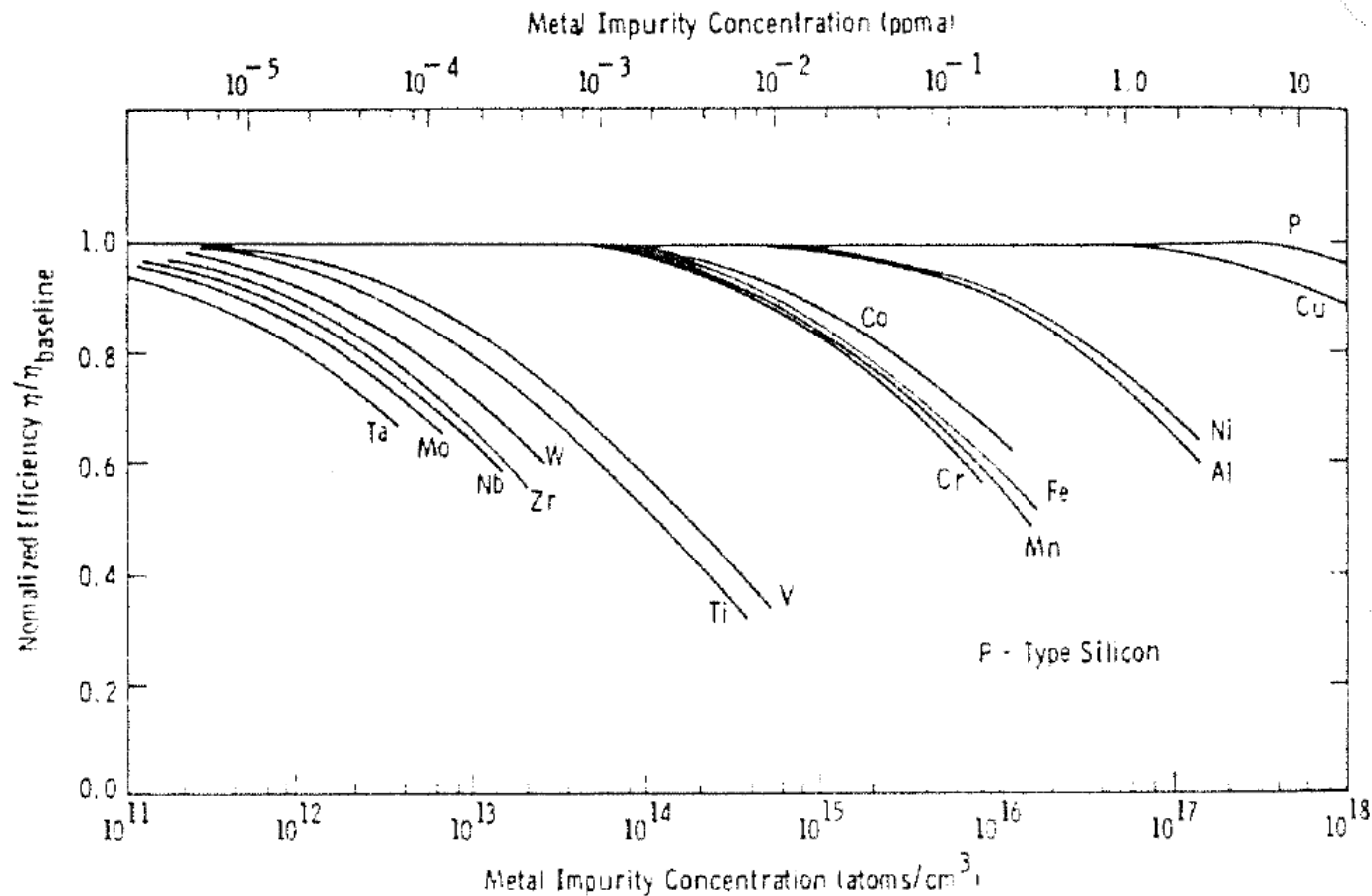


Fig. 4. Solar-cell efficiency versus impurity concentration for 4- $\Omega \cdot \text{cm}$ p-base devices.

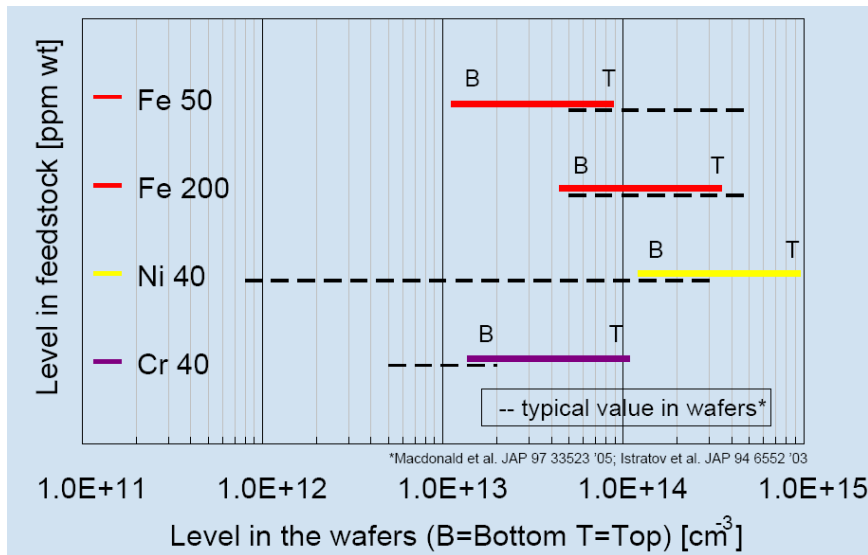
J.R. Davis *et al.*, IEEE Trans. Electron. Dev. ED-27 (1980) 677



Recent studies

Main question: Which concentrations can be allowed in the feedstock with today's crystallisation and cell processes?

Two projects (Crystal Clear and Solarfocus) produced several intentionally contaminated Ingots:



About 10 ppmw of Fe, Cr or Ni will be allowed

Effect of metals in silicon feedstock on multicrystalline silicon solar cells
Gianluca Colletti, Crystal Clear Workshop: Arriving at well-founded SoG silicon feedstock specifications, Amsterdam, 2008

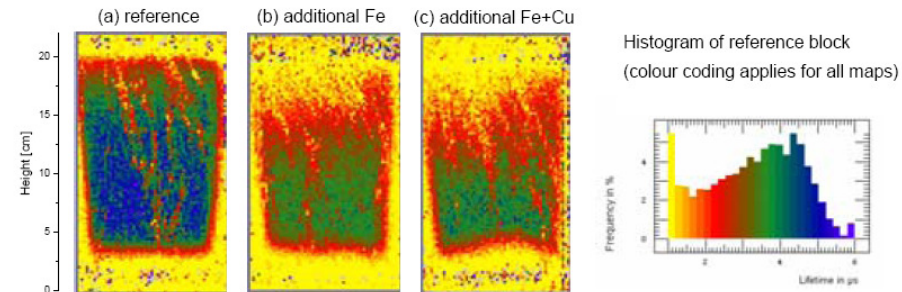
Conclusion from Projects:

→ The acceptable concentration of metals is in the ppm - range

Study of impurities in block Silicon by intentional contamination

Crystallization of small mc-Si blocks with addition of impurities to the feedstock:

- (a) reference block with high purity feedstock
- (b) addition of Fe (target concentration of 2ppma in the melt)
- (c) addition of Fe and Cu (target conc. of 2ppma and 20ppma, respectively)

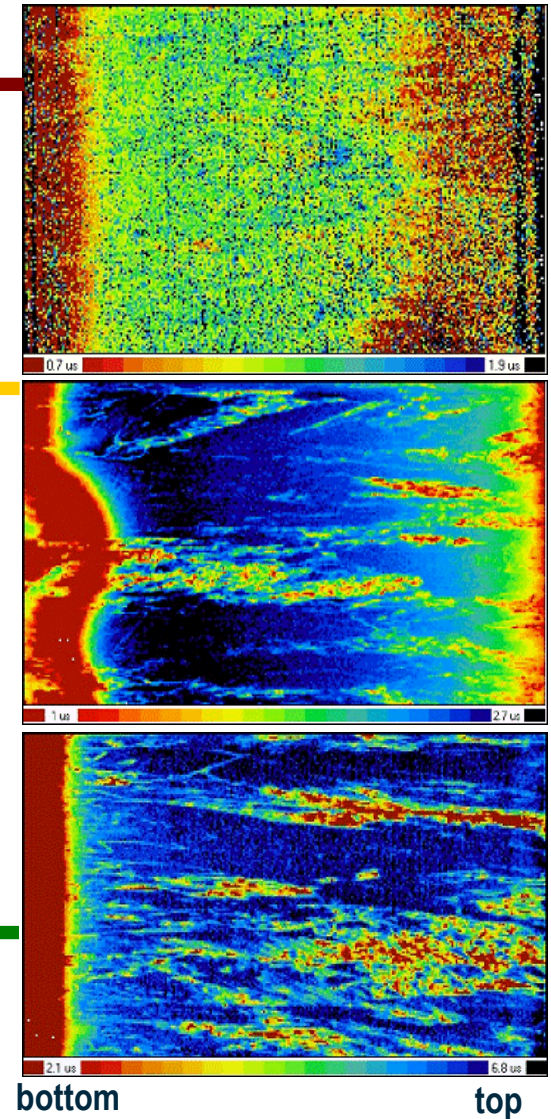
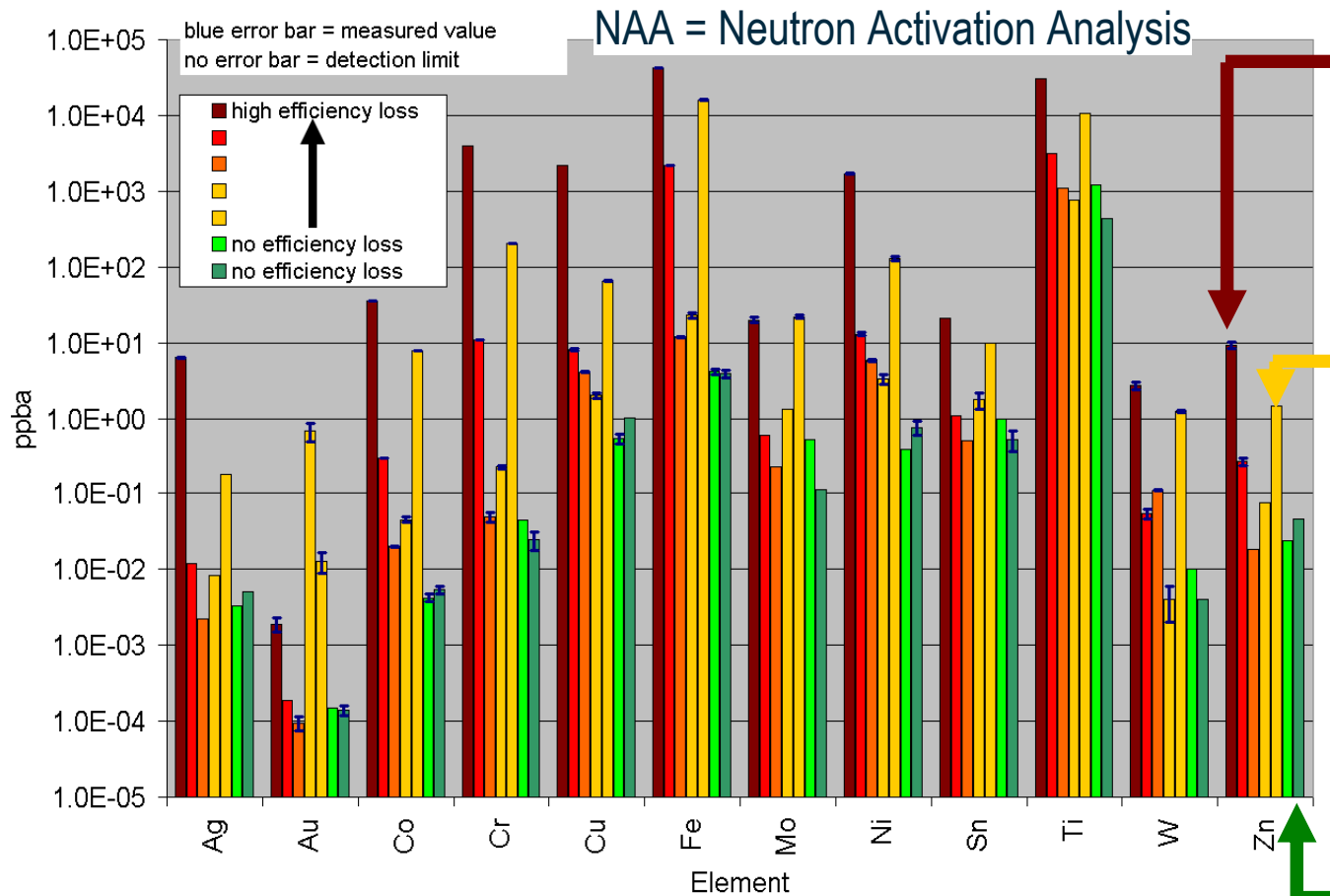


MW-PCD images of transverse section of blocks, measured at unpassivated bricks with 12.5cm width

SOLAR SILICON MATERIAL RESEARCH NETWORK – SOLARFOCUS
Stephan Riepe, PVSEC, Valencia, 2008



Results obtained with umg-Si at QC



→ Limit for metal should be below ppm range
 Today's acceptable umg-Si Qualities have metal concentrations below critical limit
 - producers have crystallisation steps in their processes
 - metals have low segregation coefficients k



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Alkaline Metals



Figure 2a



Figure 2b



Figure 2c

Figure 2: Pictures of fused silica crucible containing Si feedstock before (a) and after test (b). Crucible cover internal surface exhibiting important flaking (c).

C. Martin, CSSC3, SINTEF/NTNU Trondheim 2009

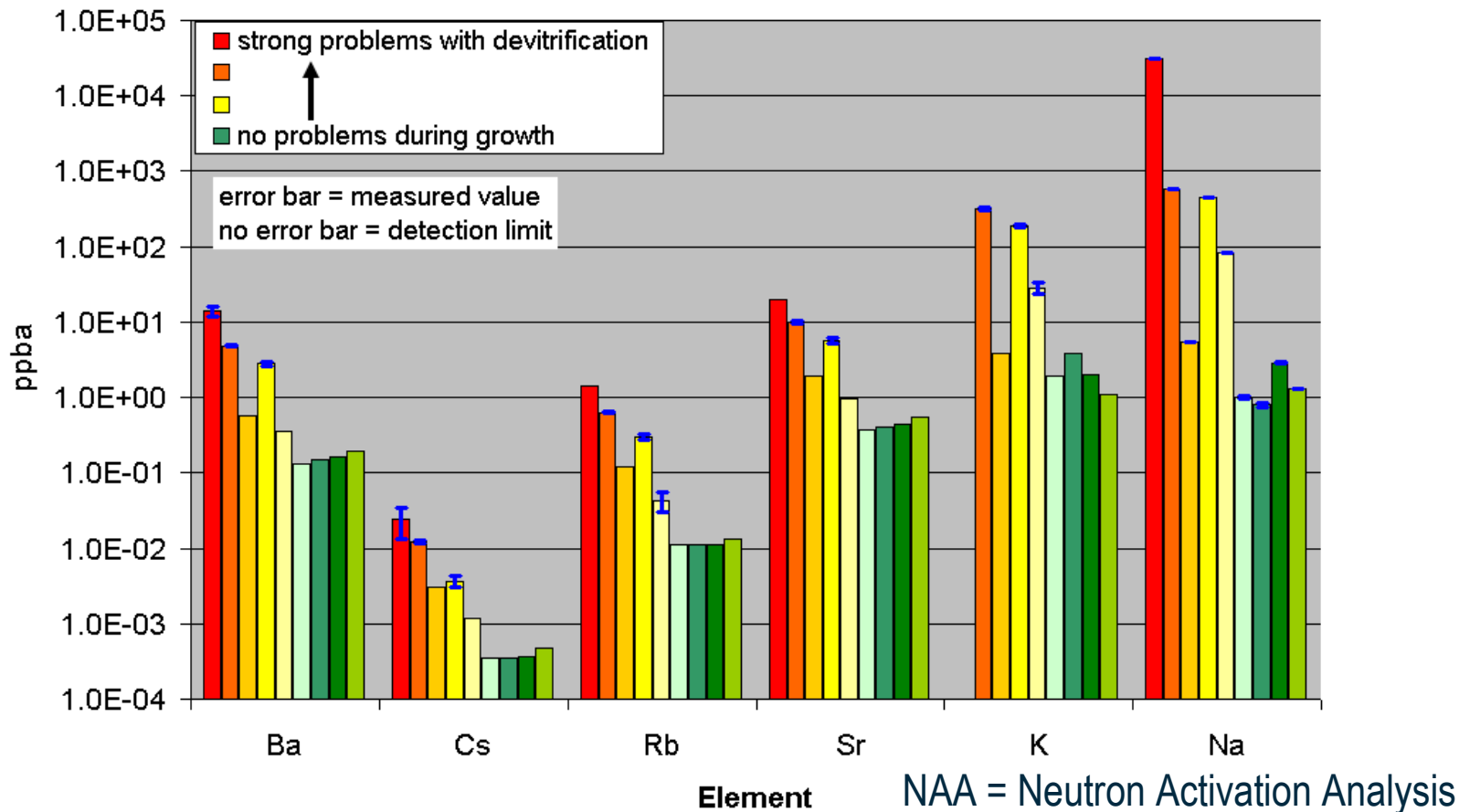
Table 1: Sodium content by AAS and cristobalite content in XRD analyses of the surface compared to the bulk of the material.

	Bulk	Surface
Na content / ppm	25	480
Cristobalite content / %	10	60



Results obtained with umg-Si at QC

Alkaline metals can lead to a devitrification („Entglasung“ / „crystallisation“) of the Quartz Crucibles → they become brittle and may break





Impurities in Silicon

Typical influences of impurities in Silicon

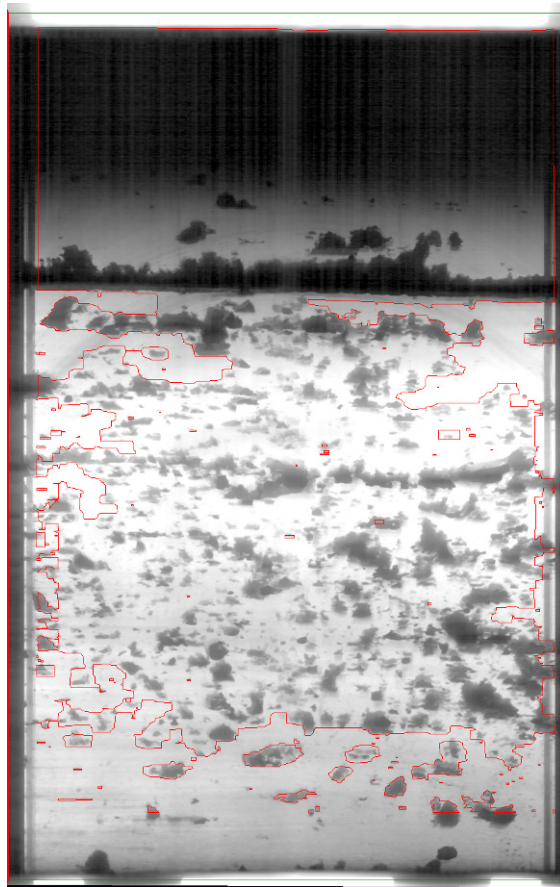
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Inclusions in Silicon Feedstock

Too high concentrations of C / N in the feedstock can lead to the formation of SiC / SiN inclusions in the ingot

IR transmission images of crystallised blocks:



Feedstock with too much C / N
→ SiC / SiN inclusions

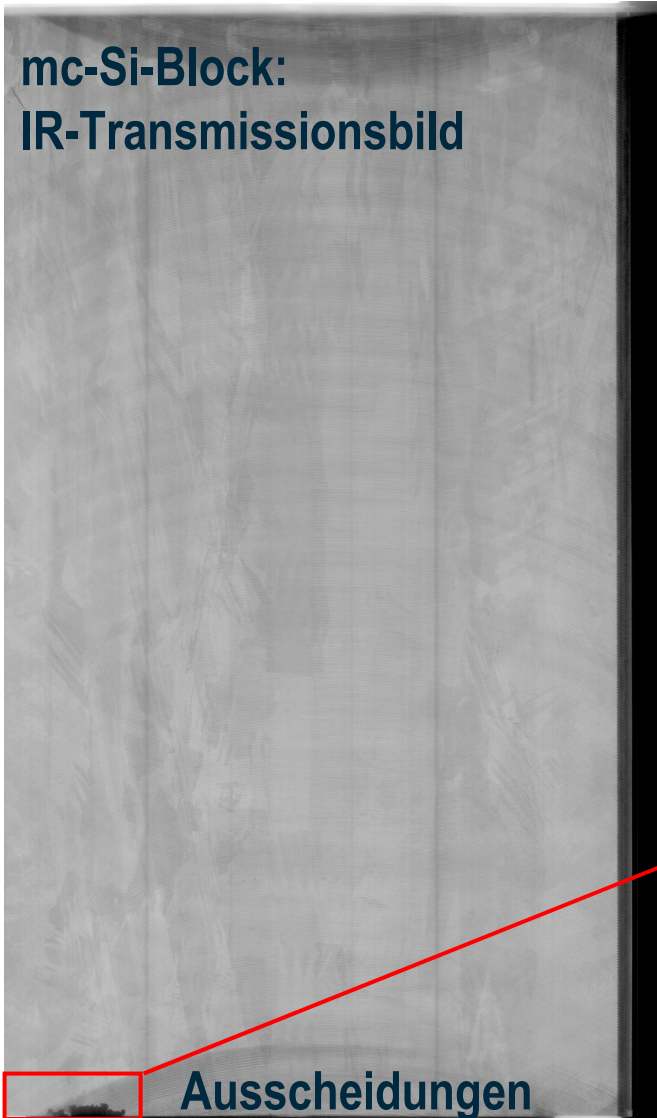
Feedstock with acceptable concentrations of C / N
→ Inclusion free block



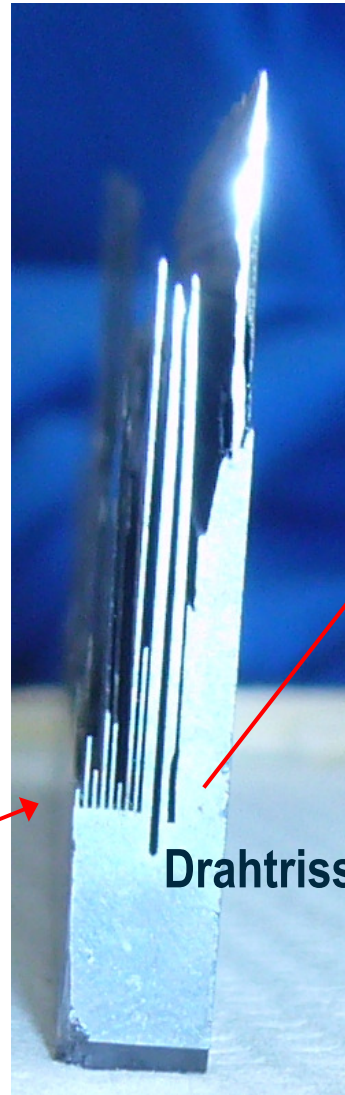


Wire breakage due to SiC inclusions

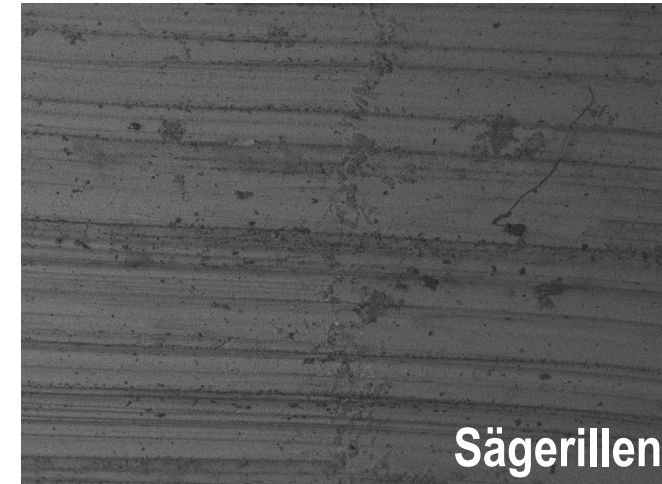
mc-Si-Block:
IR-Transmissionsbild



Ausscheidungen



Drahtriss



Sägerillen



REM + EDX: SiC, SiN



Bor-Sauerstoff-Defect → Light Induced Degradation

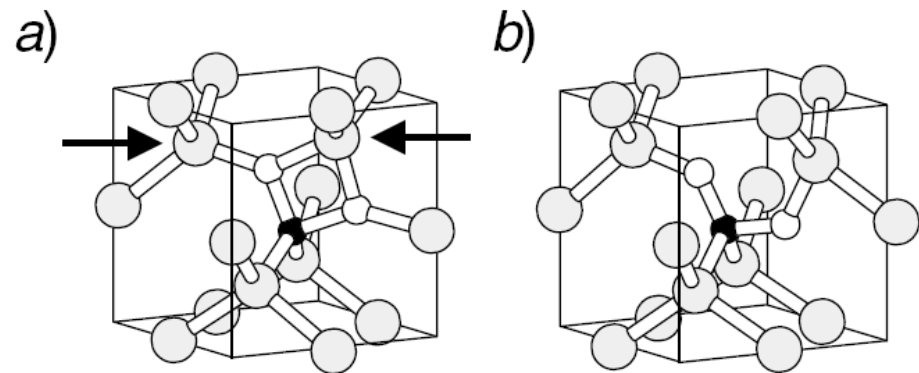
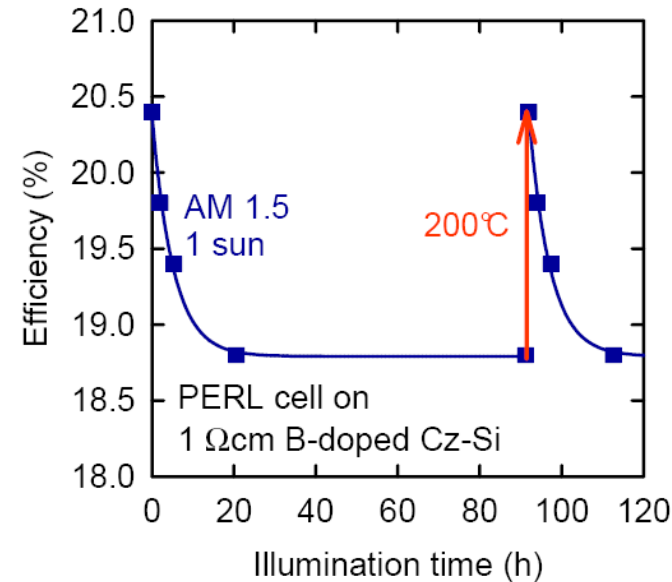
Boron doped Silicon:

- $[O_i] < 1 \times 10^{17} \text{ cm}^{-3}$ up to $> 10^{18} \text{ cm}^{-3}$
- Light induced Degradation (LID) of Efficiency

Recombination active defect density

- $N_{\text{def}} \sim [B]$
- $N_{\text{def}} \sim [O]^2$

→ Defect model with one boron and two oxygen atoms

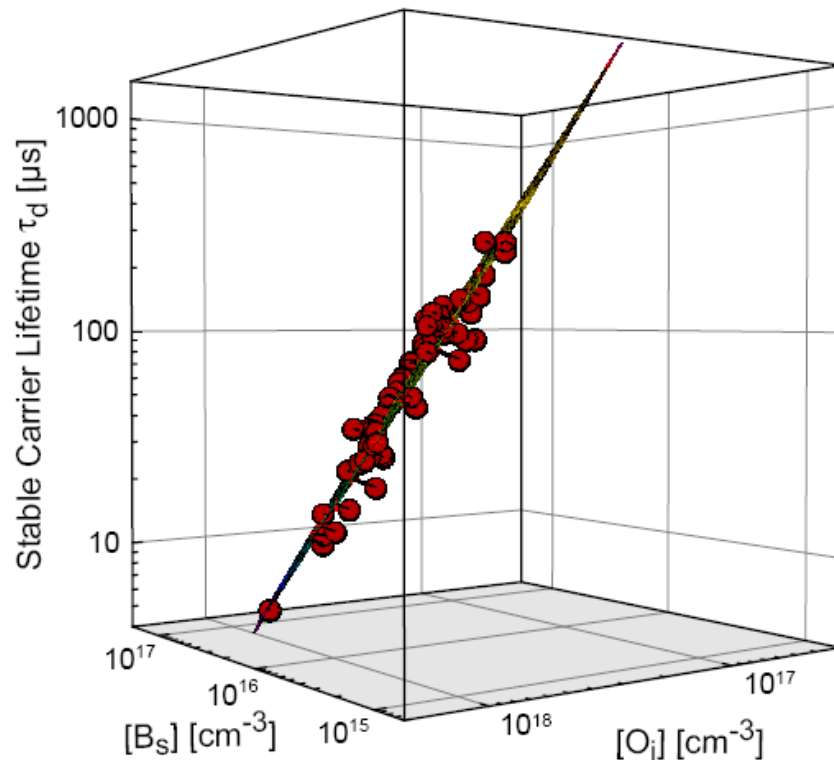


Adey *et al.*, Phys. Rev. Lett., **93** (5), (2004).



Lifetime after LID

Simultaneous fit of τ_d for varying boron and oxygen content



Remarkable small deviation between data points and fit

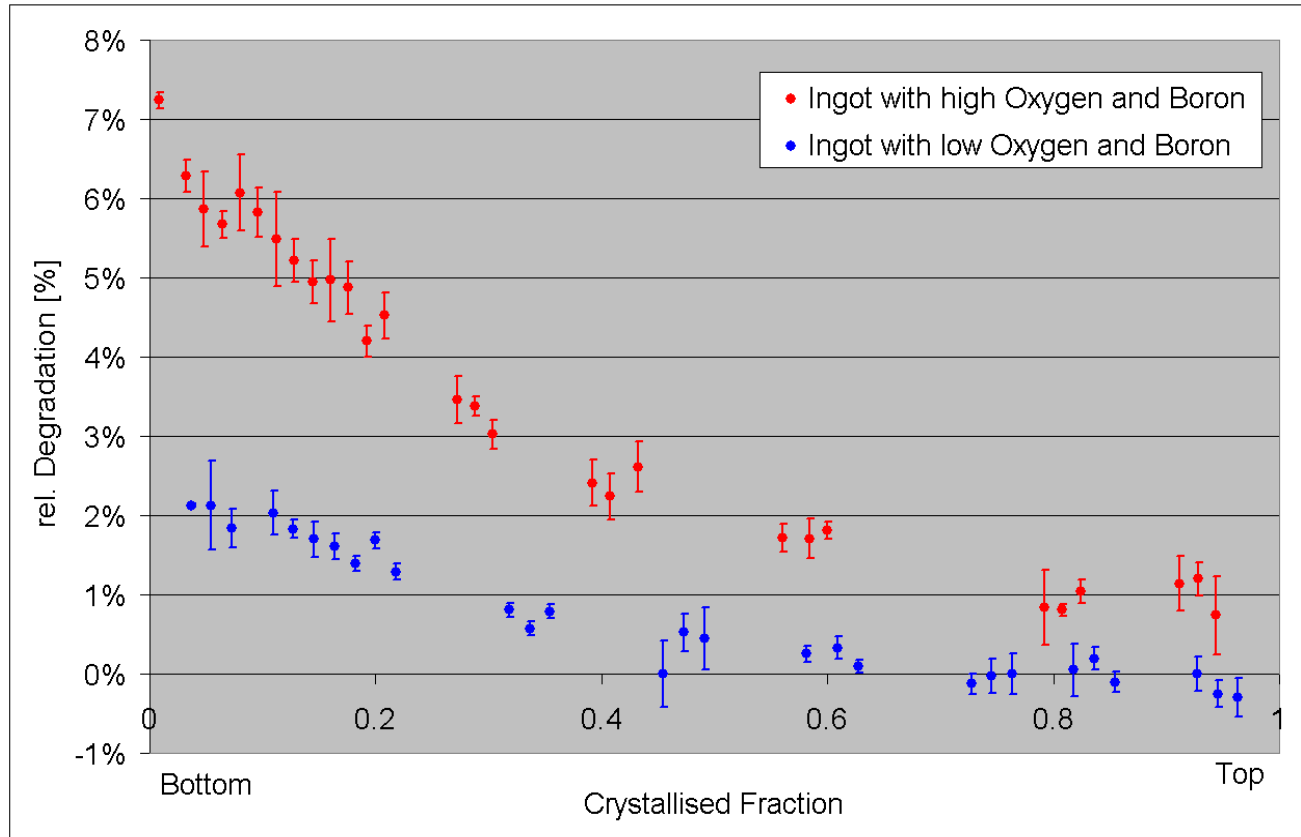
Boron and oxygen content predicts carrier lifetime !

$$\tau_d = 7.675 \times 10^{45} \cdot [B_s]^{-0.824} \cdot [O_i]^{-1.748}$$

Source: K. Bothe



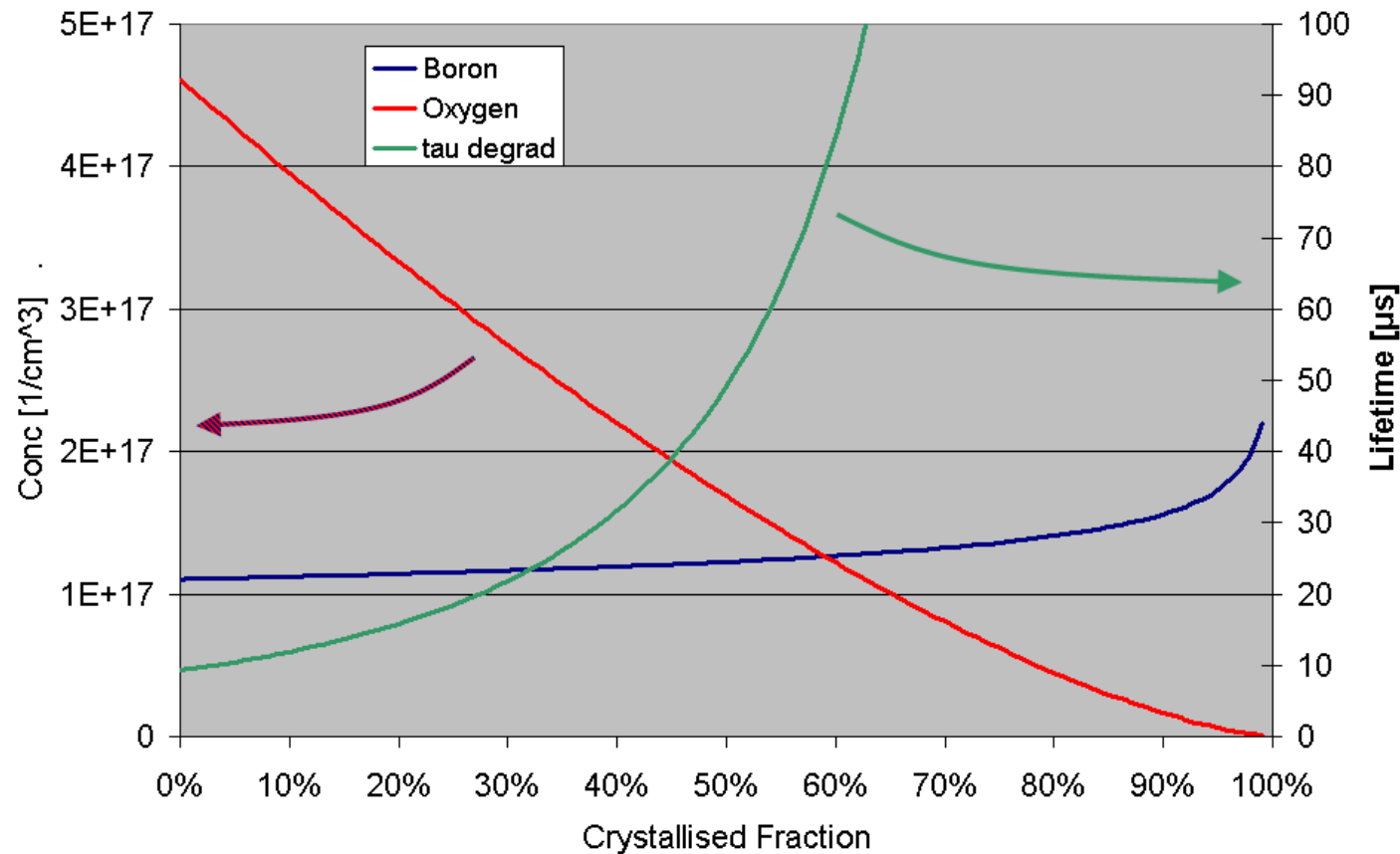
Distribution of LID for umg-Si



Using umg-Si with too high oxygen and boron concentrations can lead to very high degradation > 5% relative
Control of B and O can reduce Degradation to below 2%



Distribution of LID for umg-Si



Distribution of Degradation in Ingot can be explained well by distribution of B and O in the Ingot



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Summary

- Difference in impurity concentrations between semicond.-grade Si and umg-Si ~ three orders of magnitude (ppb vs. ppm)
- Cleaning process for metals mainly done by crystallisation (segregation)
- Cleaning of B / P main task and special know how of umg-Si producers
- High concentrations of B / P can lead to efficiency and yield losses
- High concentrations of metals can lead to additional losses of efficiency
- High C / N concentrations can lead to the formation of inclusions
- High concentrations of O can lead to strong light induced degradation



Conclusions

- **Today's best umg-Si qualities can be used to produce solar cells with acceptable and controllable properties**
 - Dopant concentrations are low enough to give reasonable efficiency and yield (still improvements are needed in future)
 - Metal concentrations are low enough and lead to no additional efficiency loss
 - C/N concentrations are low enough to make processing and sawing unproblematic
 - O concentrations low enough to give controllable low degradation (LID)
- **Still strong quality variations exist between different umg-Si suppliers**
- **umg-Si is a very attractive scientific topic with a lot of „low hanging fruits“ → fast progress was made and will probably be made in the near future in all steps of value chain**
- **now: umg-Si is an attractive second source for wafer and cell manufacturers**
- **in the future: umg-Si will be necessary to provide enough Si for Terawatt market**

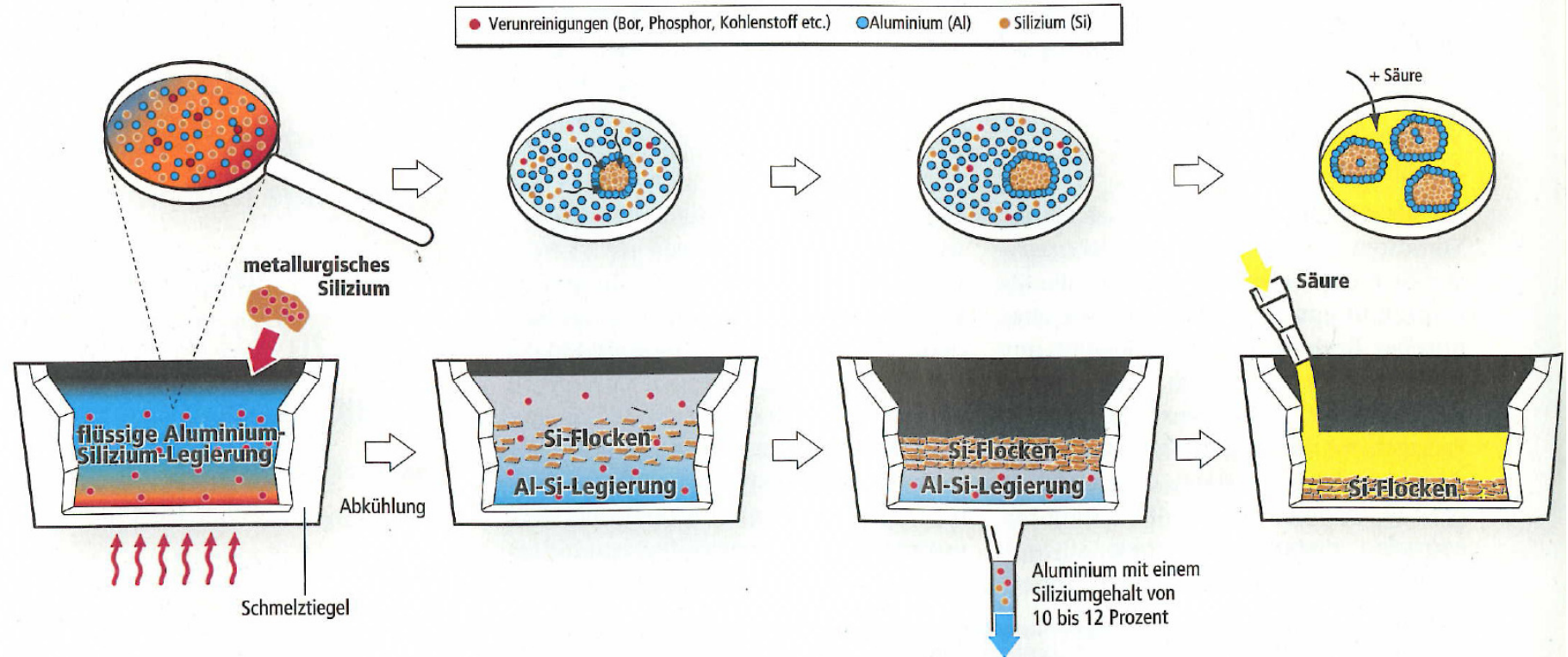


VIELEN DANK

Q.CELLS



Process 6N Silicon



In einer Aluminiumschmelze mit einer Temperatur von rund 800 Grad Celsius wird metallurgisches Silizium aufgelöst, das mit chemischen Elementen wie Bor oder Phosphor verunreinigt ist. Dieses besitzt eine Reinheit von 97 Prozent oder höher.

Beim Abkühlen der flüssigen Aluminium-Silizium-Legierung kristallisiert zuerst das Silizium in Form von Flocken aus. Die Verunreinigungen bleiben dabei im Aluminium zurück, was den primären Reinigungsschritt des Verfahrens darstellt. Um die Reinheit der Siliziumflocken zu erhöhen, werden weitere metallurgische Reinigungsschritte wie etwa Ausgasungen vorgenommen.

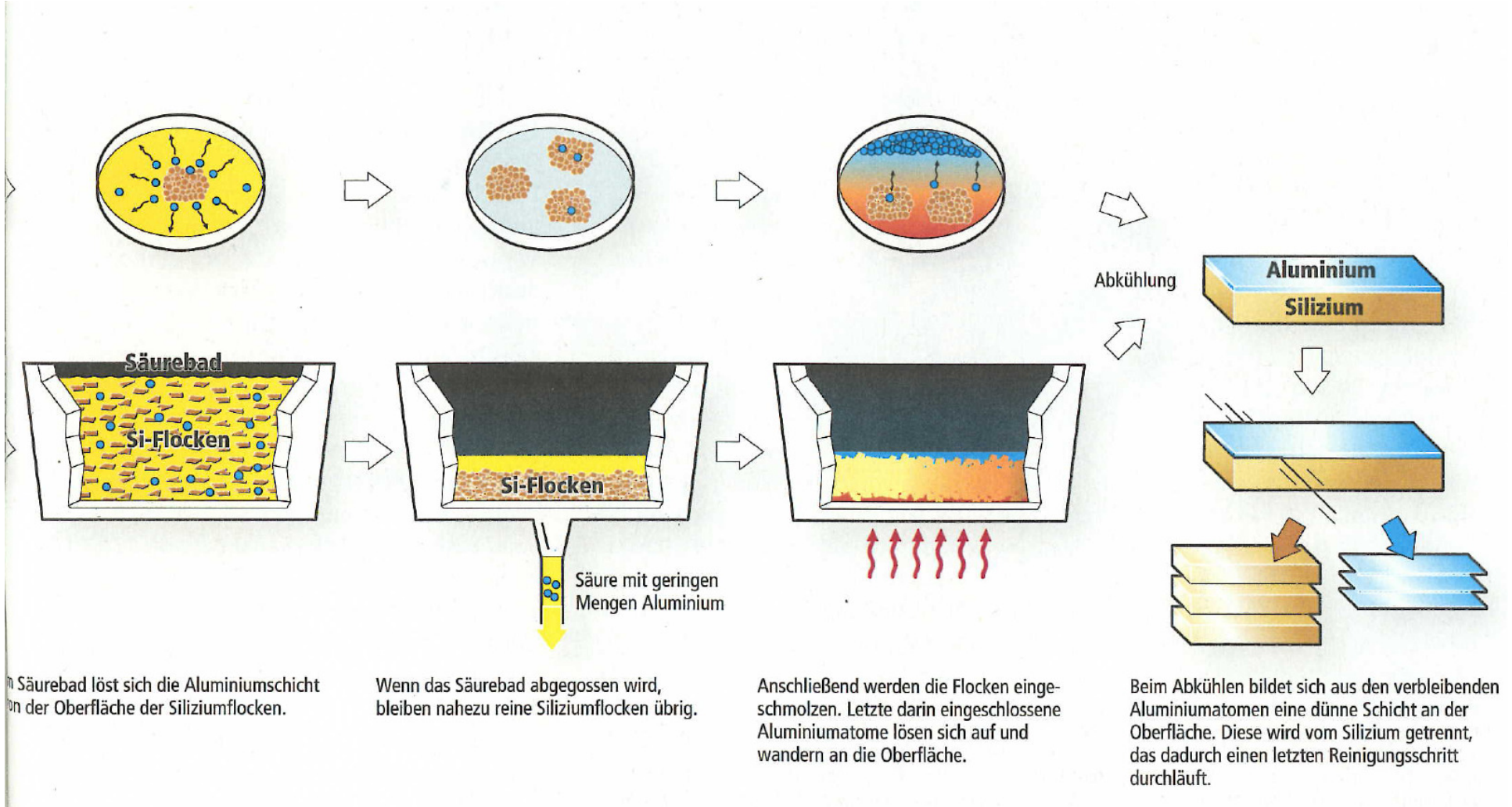
Ist die Auskristallisierung des Siliziums abgeschlossen, wird das flüssige Aluminium abgossen. Da es mit 10 bis 12 Prozent Silizium angereichert ist, wodurch seine Stabilität erhöht wird, kann es gewinnbringend verkauft werden.

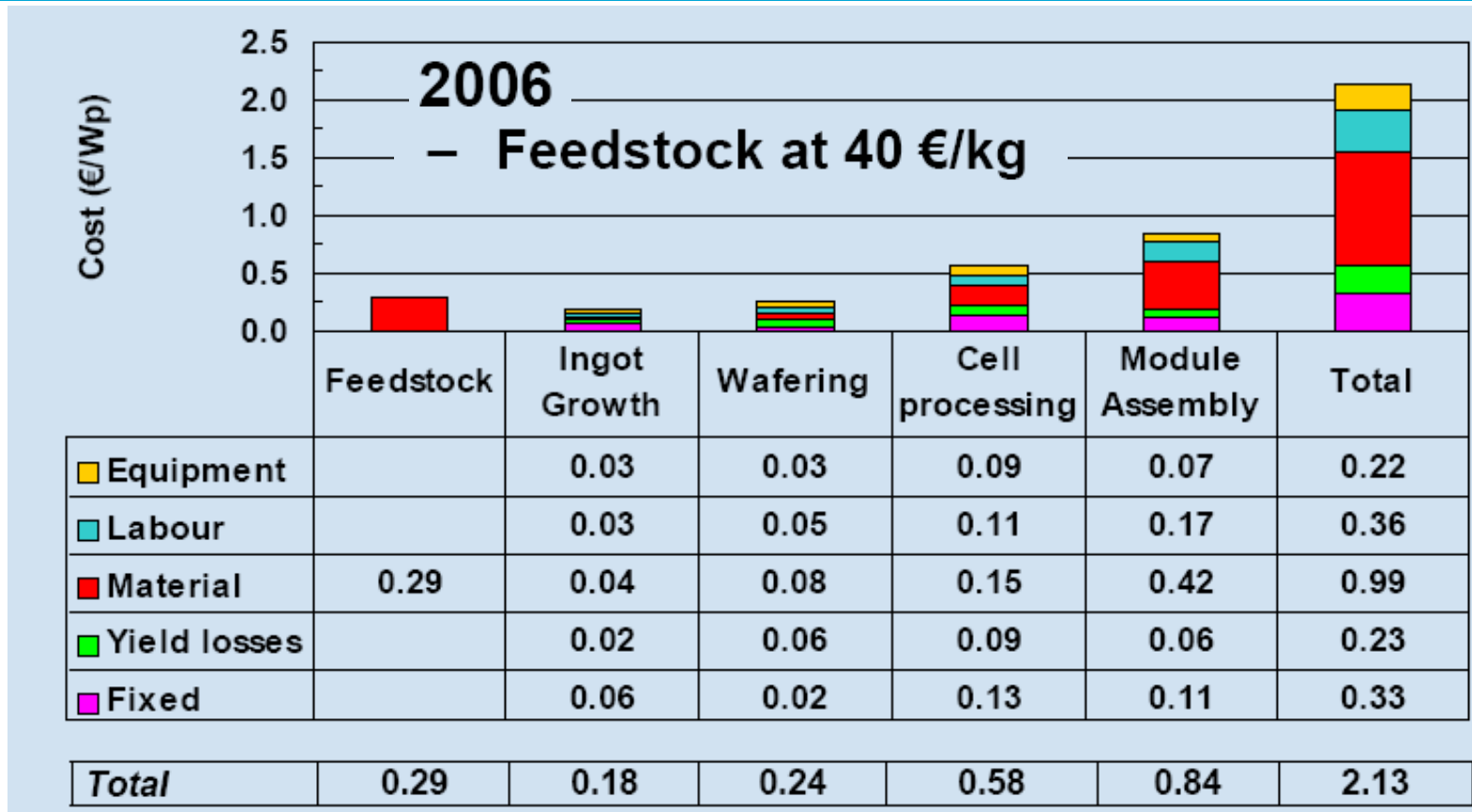
Die Siliziumflocken sind noch mit einer dünnen Schicht Aluminium überzogen. Um diese zu entfernen, wird das Silizium mit Säure übergossen.

Source: Photon



Process 6N Silicon



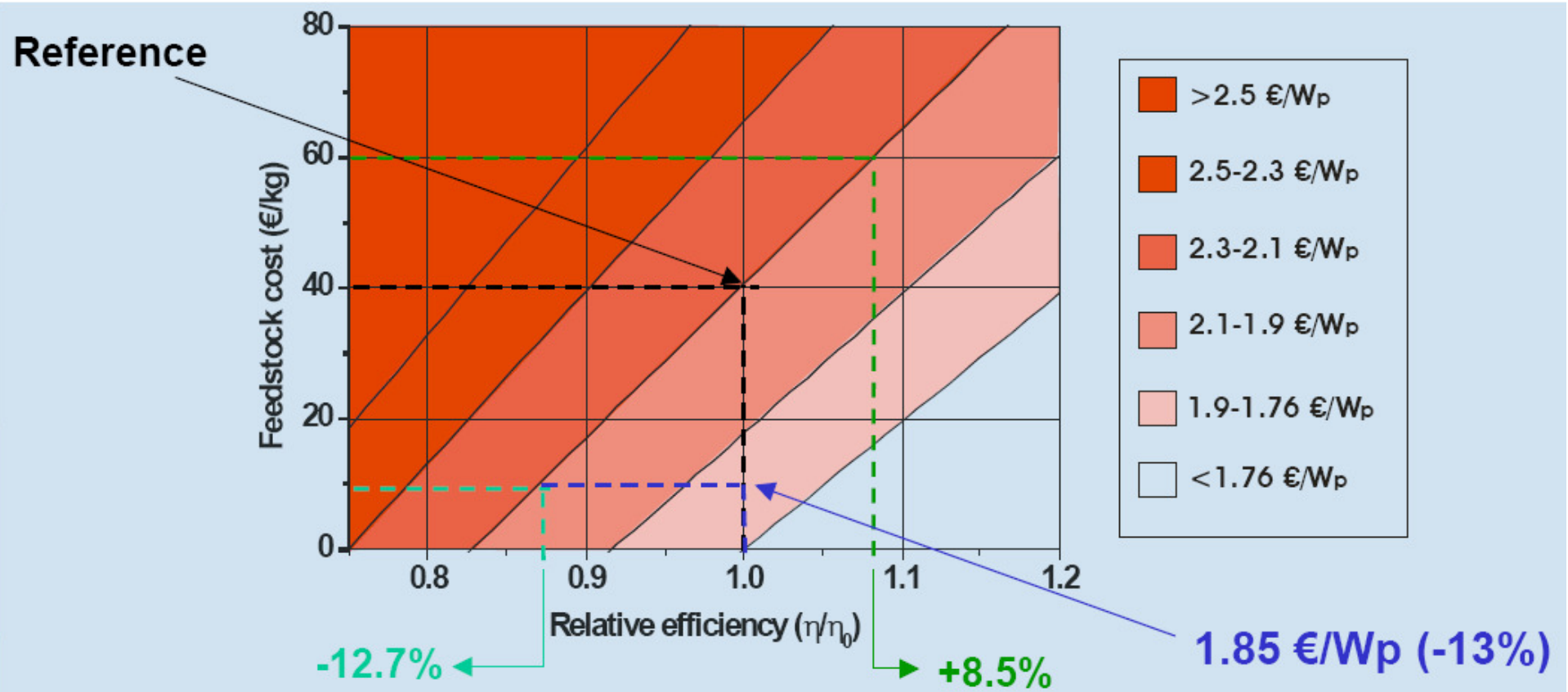


Impact of Si feedstock features on PV costs, Carlos del Cañizo (UPM – IES), Crystal Clear Workshop: Arriving at well-founded SoG silicon feedstock specifications, Amsterdam, 2008

In 2007 / 2008: Silicon prices on the spot market went up to 400 \$ / kg

→ Feedstock costs were up to 50% of module costs

Situation now: Silicon prices have gone down to ~ 40€/kg again



Impact of Si feedstock features on PV costs, Carlos del Cañizo (UPM – IES), Crystal Clear Workshop: Arriving at well-founded SoG silicon feedstock specifications, Amsterdam, 2008

Situation now: Silicon prices have gone down to ~ 40€/kg again
Still umg-Si can be a commercially attractive material if price is significantly lower
But: only very small efficiency losses are allowed !